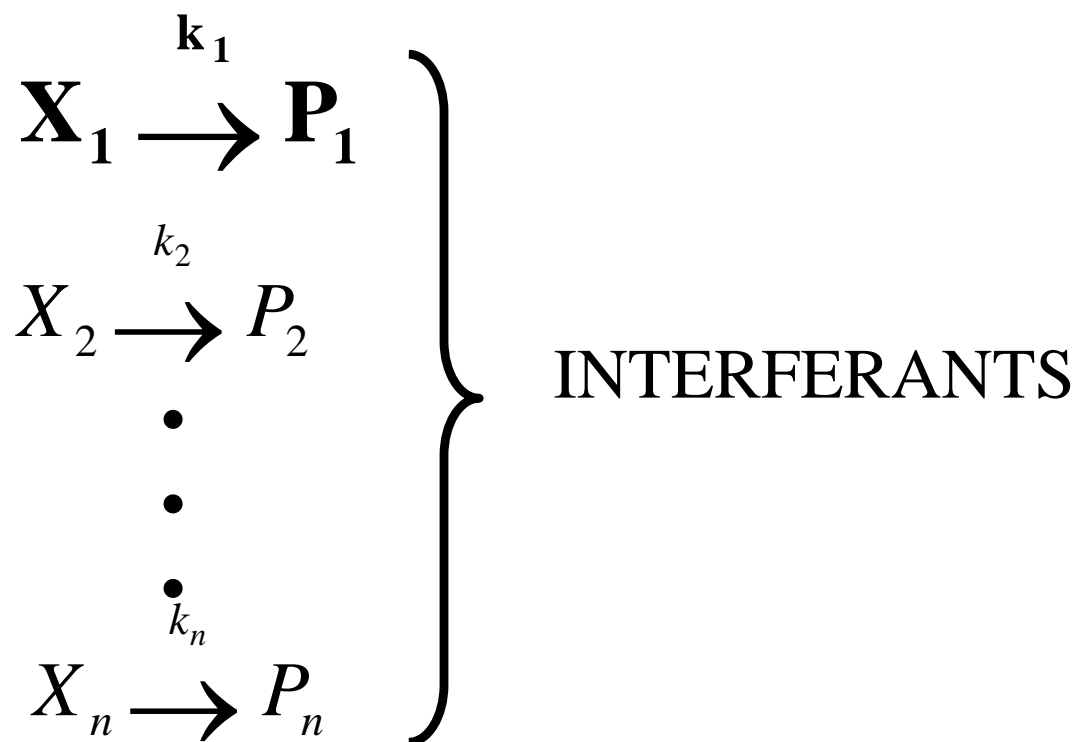
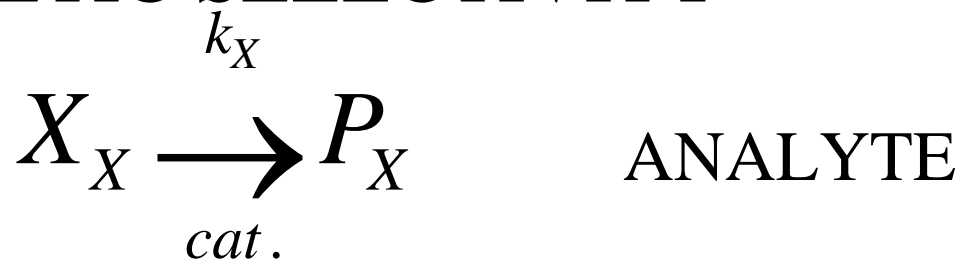


ORIGINS OF SELECTIVITY. PART II

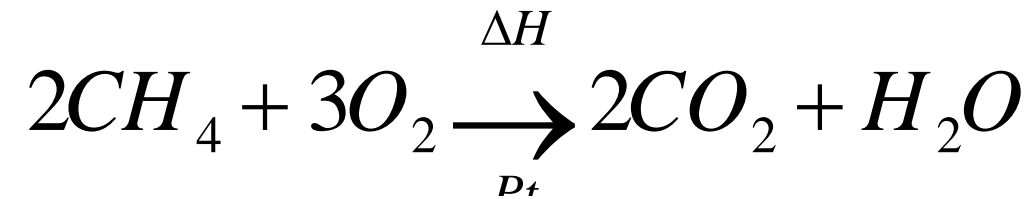
KINETIC SELECTIVITY



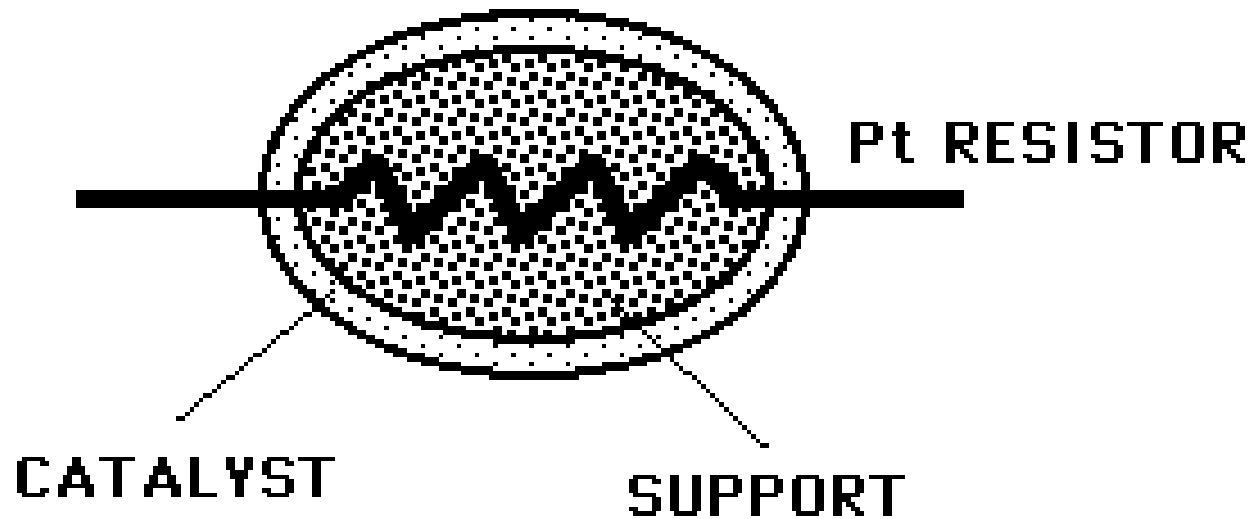
$$k_X \gg k_1, k_2 \dots k_n$$

... USE CATALYSTS

PELLISTOR

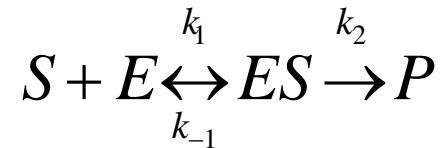


0.5–1 mm



ENZYME SELECTIVITY

- ENZYMES ARE HIGHLY VERSATILE CATALYSTS



- AT STEADY STATE

$$\frac{dC_{ES}}{dt} = k_1 C_S C_E - (k_{-1} + k_2) C_{ES} = 0$$

- MICHAELIS MENTEN

$$v = \frac{v_{\max} C_S}{C_S + K_m}$$

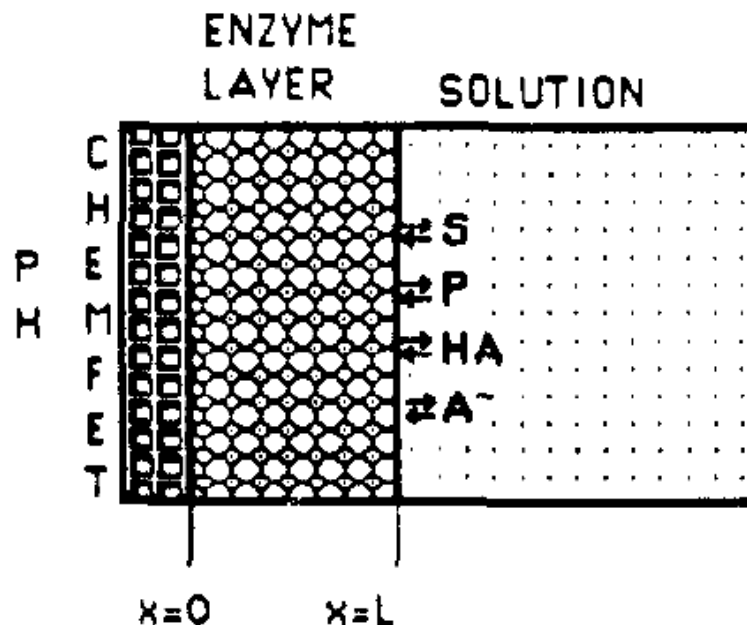
Michaelis-Menten constant K_m is figure of merit of the enzyme

M-M KINETICS

- ENZYMATIC REACTIONS ARE pH DEPENDENT

$$v = \mathfrak{R}_{pH} \frac{v_{\max} C_S}{C_S + K_m} \quad \mathfrak{R}_{pH} = \left(1 + \frac{C_H}{K_{ES_1}} + \frac{K_{ES_2}}{C_H} \right)$$

- ENZYMATIC SELECTIVITY CAN BE USED WITH ANY TRANSDUCTION PRINCIPLE EXCEPT MASS SENSING



DIFFUSION-REACTION
MECHANISM

DIFFUSION - REACTION MECHANISM

- FOR UNI-DIRECTIONAL TRANSPORT CHANGE OF C_s AT $X=0$ IS

$$\frac{\delta C_s}{\delta t} = D_s \frac{\delta^2 C_s}{\delta x^2} - \frac{\mathfrak{R}_{pH} v_{\max} C_s}{(C_s + K_m)}$$

- AFTER NORMALIZATION

$$\left(\frac{\delta C_s}{\delta t} \right)^* = D_s \left(\frac{\delta C_s}{\delta x^2} \right)^* - \phi^2 \left(\frac{C_s}{1 + C_s} \right)^*$$

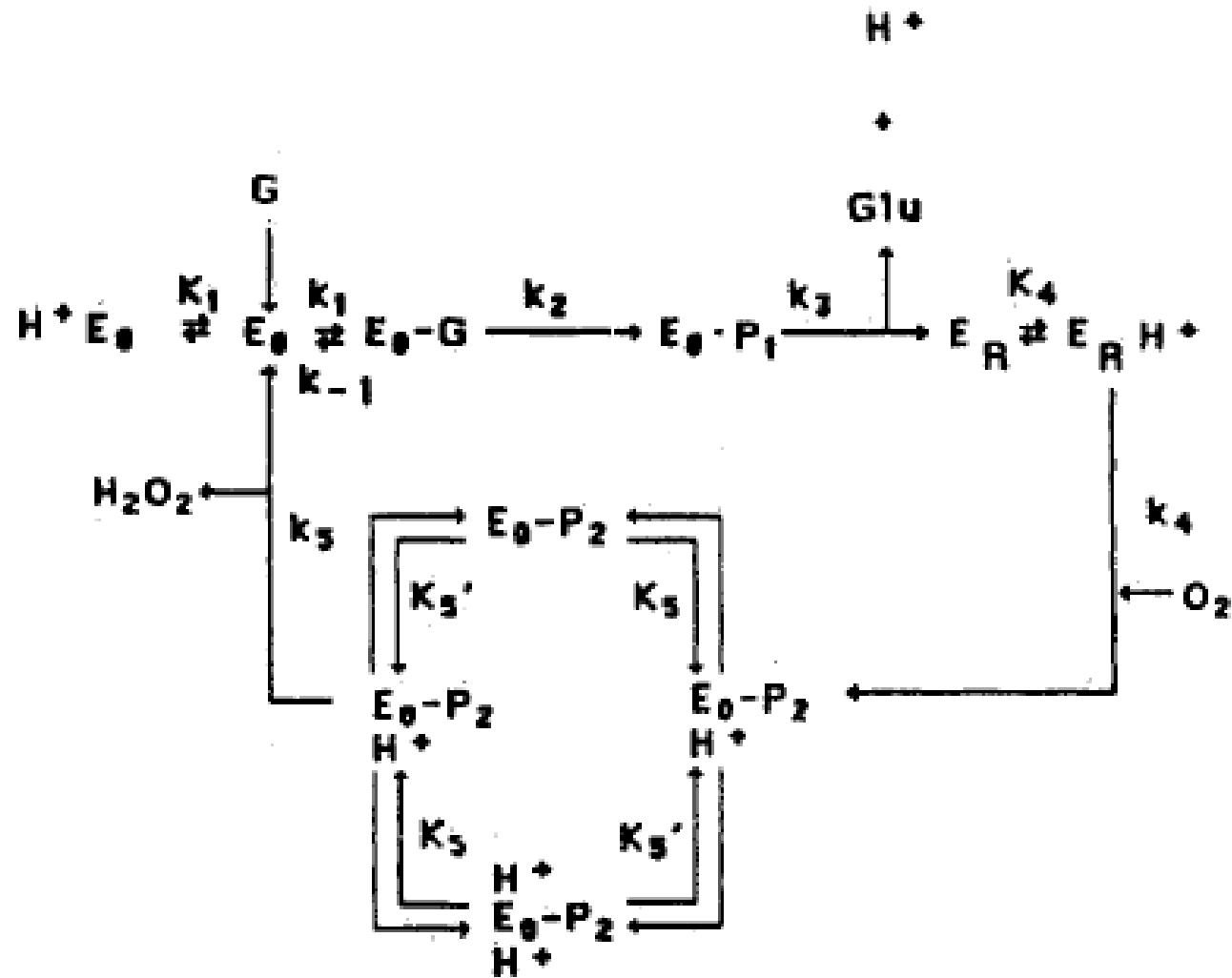
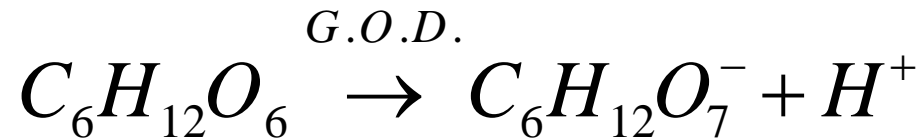
THIELE MODULUS:

$$\phi = \frac{L v_{\max}^{1/2}}{(K_m D_s \mathfrak{R}_{pH})^{1/2}}$$

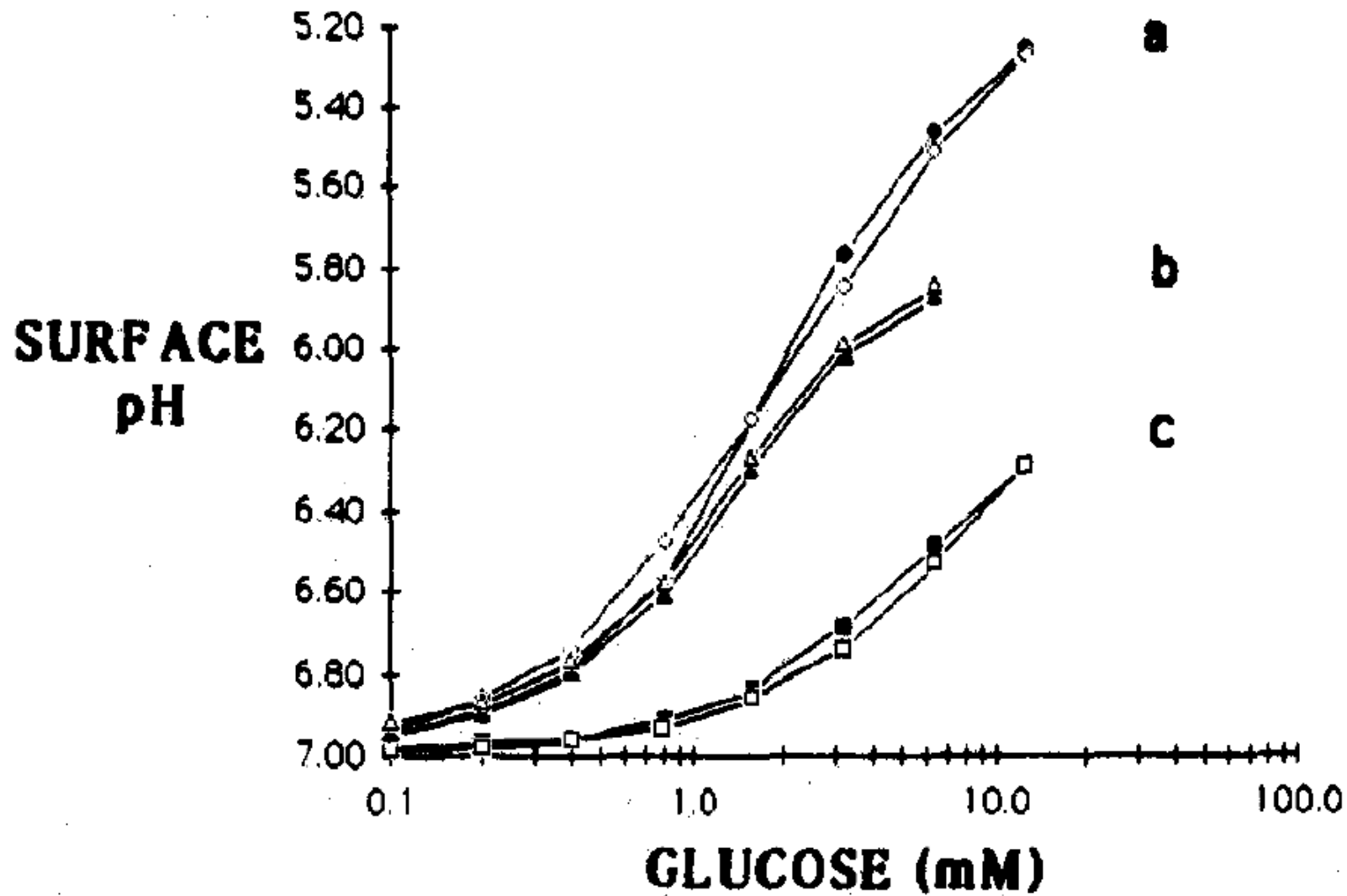
(diffusion control) $10 < \phi < 5$ (reaction control)

EXAMPLE

GLUCOSE SENSOR

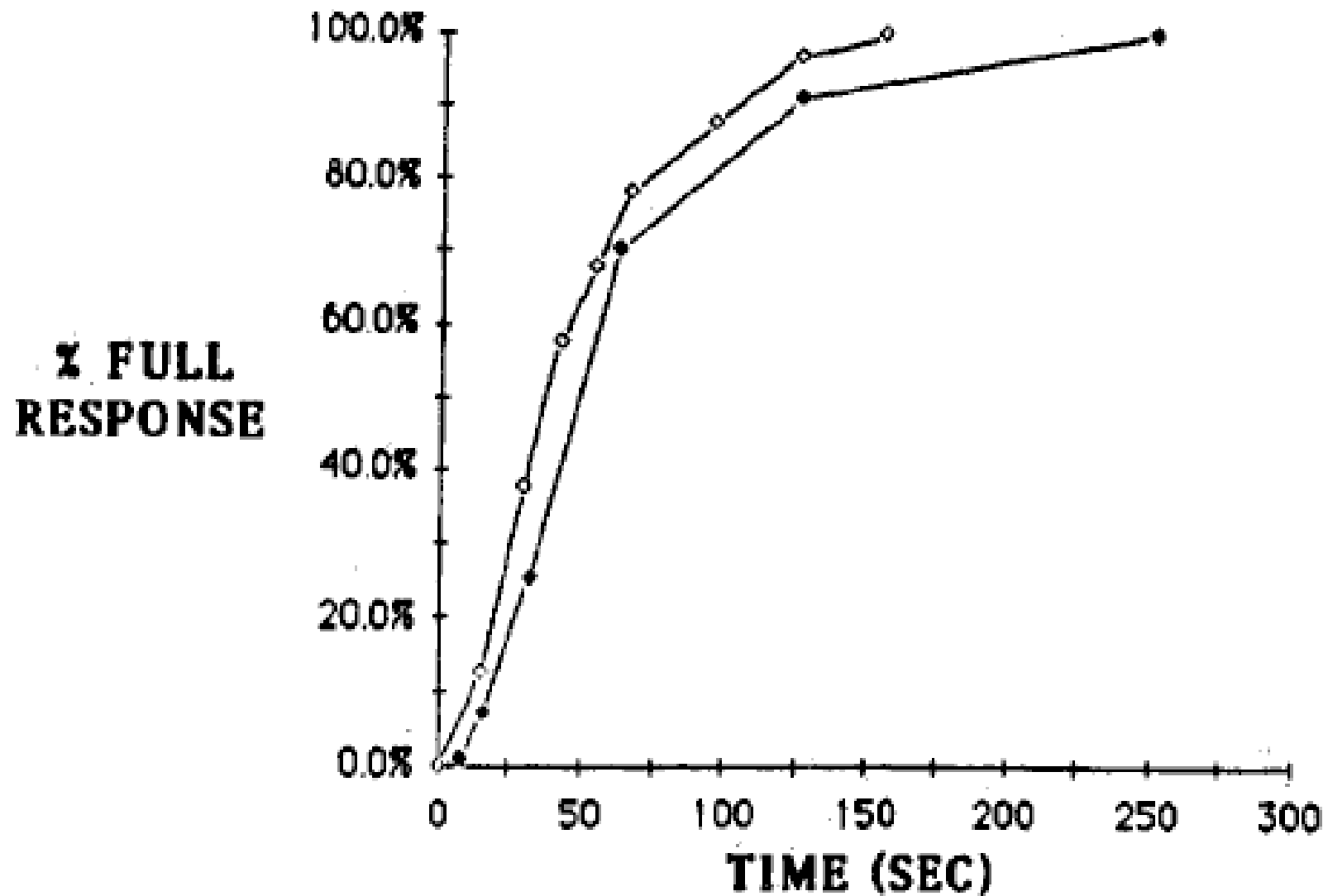


RESPONSE CURVES

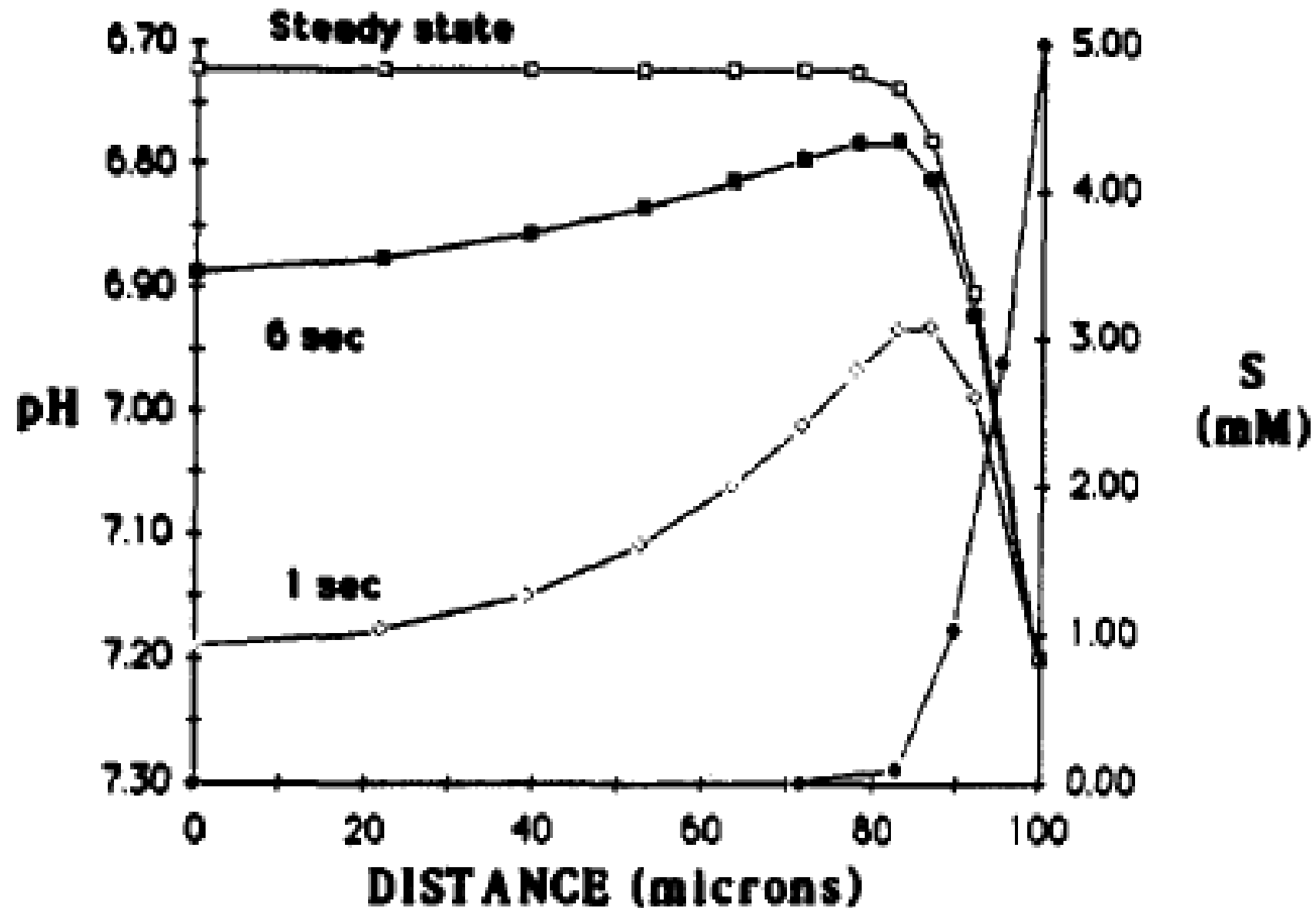


TIME RESPONSE CURVES

GLUCOSE SENSOR

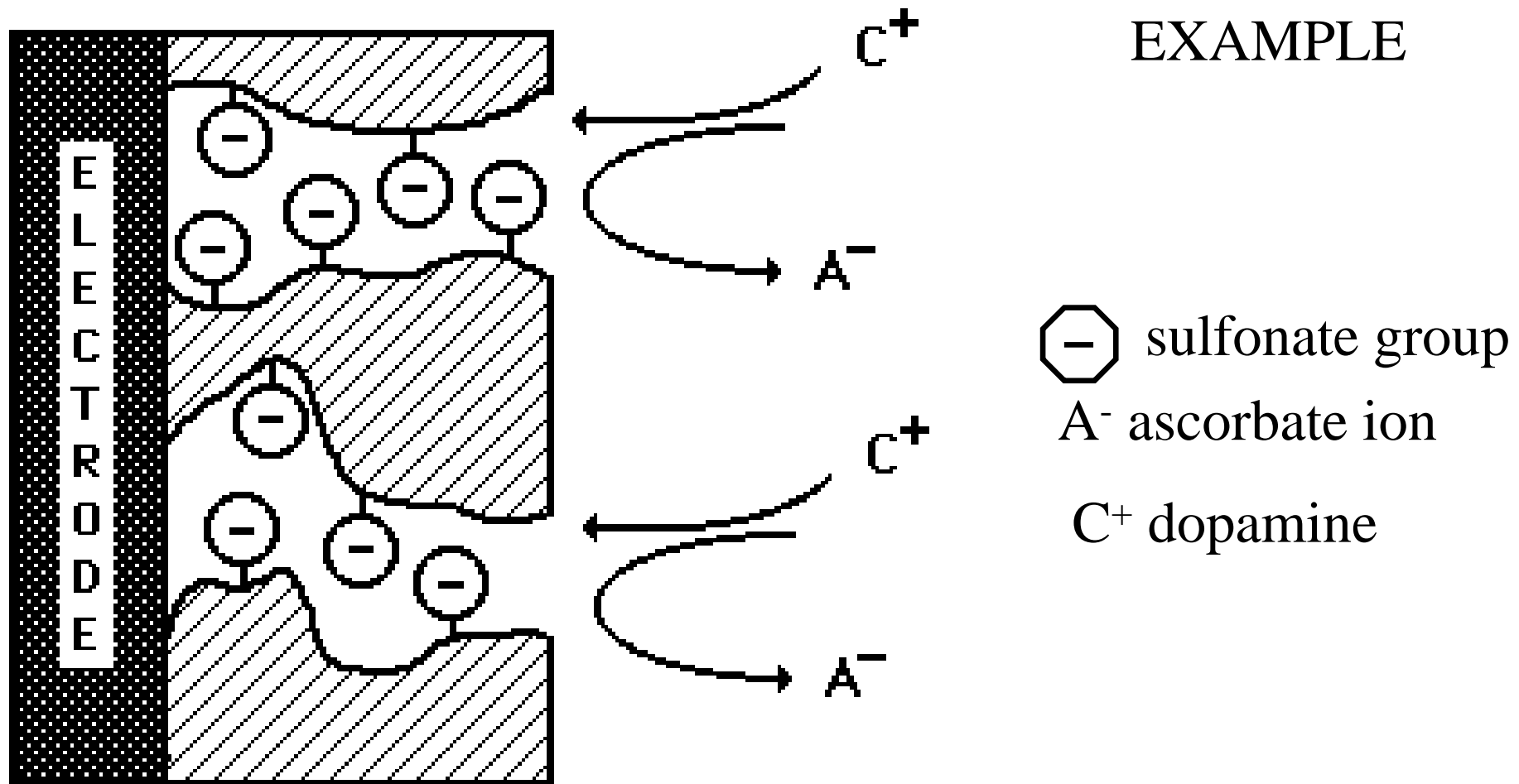


CONCENTRATION PROFILES



MASS TRANSPORT LIMITED SELECTIVITY

ELECTROSTATIC REJECTION



Higher Order Chemical Sensors

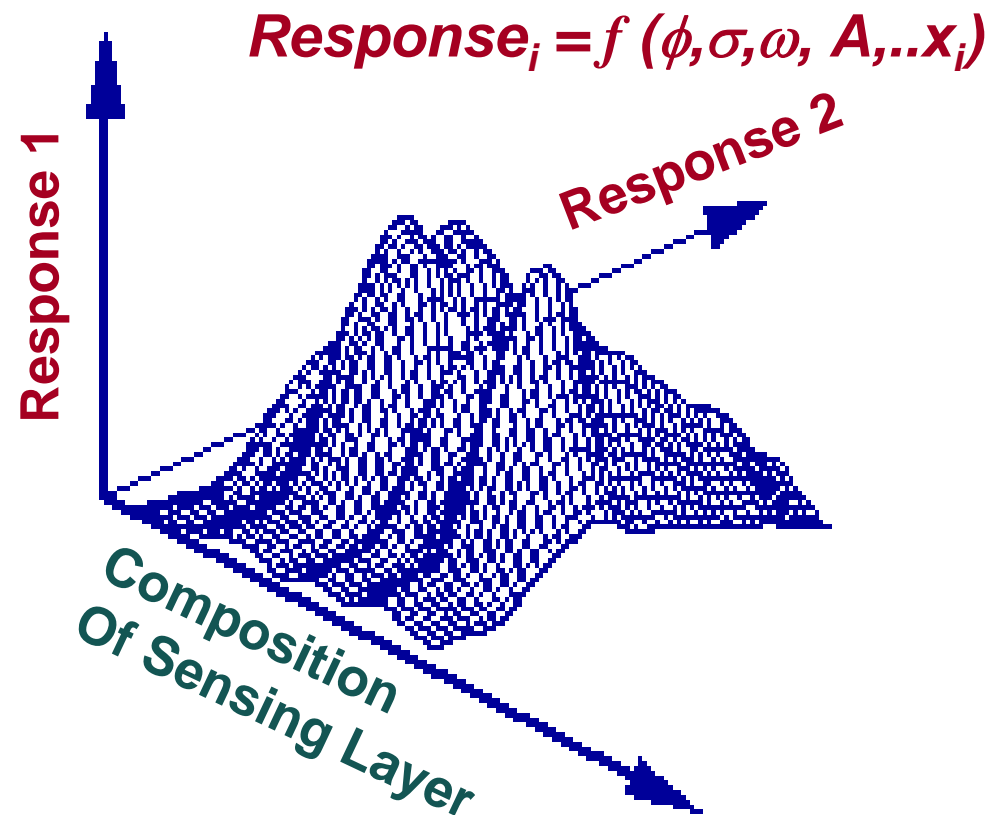
Extending the Feature Space in Chemical Sensing

ENVIRONMENTAL

- detection limit
- selectivity
- stability
- dynamic range

PROCESS CONTROL

- speed
- robustness
- safety
- stability



**Cost of obtaining the information
must not be higher than the consequences of not having that information**