FT-IR Reflection Techniques

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Overview - Main Principles of Reflection Techniques
Internal Reflection
External Reflection
Summary
Differences Between Transmission and Reflection FT-IR Techniques

Transmission:

• Excellent for solids, liquids and gases
• The reference method for quantitative analysis
• Sample preparation can be difficult

Reflection:

• Collect light reflected from an interface air/sample, solid/sample, liquid/sample
• Analyze liquids, solids, gels or coatings
• Minimal sample preparation
• Convenient for qualitative analysis, frequently used for quantitative analysis
FT-IR Reflection Techniques

Internal Reflection Spectroscopy:
   - Attenuated Total Reflection (ATR)

External Reflection Spectroscopy:
   - Specular Reflection (smooth surfaces)

Combination of Internal and External Reflection:
   - Diffuse Reflection (DRIFTs) (rough surfaces)
FT-IR Reflection Techniques

- Infrared beam reflects from a interface via total internal reflectance
- Sample must be in optical contact with the crystal
- Collected information is from the surface
  - Solids and powders, diluted in a IR transparent matrix if needed
  - Information provided is from the bulk matrix
- Sample must be reflective or on a reflective surface
  - Information provided is from the thin layers
Attenuated Total Reflection (ATR)

- introduced in the 1960s, now widely used
- light introduced into a suitable prism at an angle exceeding the critical angle for internal reflection ⇒ an evanescent wave at the reflecting surface

**Single Bounce ATR**

- sample in close contact with IRE
- from the interaction of the evanescent wave (exponential decay) with the sample, a spectrum can be recorded with little or no sample preparation

IRE - internal reflection element = ATR crystal
Total internal reflection

- **critical angle** - when the angle of refraction \((r)\) becomes equal to 90 degrees and Snell's law reduces to:

\[
\sin(\theta) = \frac{n(1)}{n(2)} \quad n(1) \neq n(2)
\]

where \((\theta)\) is termed the **critical angle**

When the critical angle is exceeded for a particular light wave, it exhibits **total internal reflection** back into the medium.

The larger the angle to the normal, the smaller is the fraction of light transmitted, until the angle when **total internal reflection** occurs.
Factors influencing ATR analysis

- Wavelength of IR radiation \( \lambda \)
- Refractive indexes of sample and IRE \( n_{\text{smp}}, n_{\text{IRE}} \)
- Angle of incidence of IR radiation \( \theta \)
- Depth of penetration (pathlength) \( d_p \)
- Sample and IRE contact efficiency

When an incident ray is totally internally reflected at the interface between two materials of different refractive index, the intensity of the evanescent field extending into the medium of lower index decays exponentially with distance from the boundary:

\[
l_{ev} = I_o \exp \left[ -\frac{z}{d_p} \right]
\]

- \( z \) is the distance normal to the optical interface,
- \( I_o \) is the intensity at \( z = 0 \),
- \( d_p \) is the penetration depth
**Attenuated Total Reflection (ATR)**

Depth of penetration (pathlength) of the infrared beam into the sample depends on $\lambda$, $n_{smp}$, $n_{IRE}$, $\theta$

\[
dp = \frac{\lambda}{2 \pi n_{IRE} \sqrt{\sin^2 \theta - (n_{smp} / n_{IRE})^2}}
\]

$d_p$ typically $< 10 \ \mu m$

The effective pathlength of the spectrum collected varies with the wavelength of the radiation:
- longer $\lambda \Rightarrow$ greater $d_p$: $d_p$ lower at higher wavenumbers
  - ATR intensities decreased at higher wavenumbers if compared to transmission spectra

ATR correction accounts for this variation in effective pathlength by scaling the ATR spectrum accordingly. Most FTIR software packages incorporate an ATR correction algorithm.
Attenuated Total Reflection (ATR)

**Critical Angle** depends on $n_{\text{IRE}}$ and $n_{\text{smp}}$

- increasing $n_{\text{IRE}}$ ⇒ decreasing $\theta$ and $d_p$

⇒ high values of $n_{\text{IRE}}$ needed

**Materials of ATR crystals (IRE elements)**

<table>
<thead>
<tr>
<th>MIRacle Crystal Plate</th>
<th>Application</th>
<th>Hardness kg/mm²</th>
<th>Cutoff cm⁻¹, Spectral Range</th>
<th>Refractive Index @ 1000 cm⁻¹</th>
<th>Depth of Penetration @ 45°,μ</th>
<th>pH Range of Sample</th>
</tr>
</thead>
<tbody>
<tr>
<td>AMTIR</td>
<td>Harder than ZnSe, ok with acid samples</td>
<td>170</td>
<td>630</td>
<td>2.5</td>
<td>1.70</td>
<td>1 – 9</td>
</tr>
<tr>
<td>Diamond/KRS-5</td>
<td>When you need full mid-IR spectral range</td>
<td>5700</td>
<td>250</td>
<td>2.4</td>
<td>2.00</td>
<td>1 – 14</td>
</tr>
<tr>
<td>Diamond/ZnSe</td>
<td>Ideal for hard samples, acids or alkaline</td>
<td>5700</td>
<td>525</td>
<td>2.4</td>
<td>2.00</td>
<td>1 – 14</td>
</tr>
<tr>
<td>Ge</td>
<td>General purpose and carbon filled or rubber</td>
<td>550</td>
<td>575</td>
<td>4.0</td>
<td>0.66</td>
<td>1 – 14</td>
</tr>
<tr>
<td>Si/ZnSe</td>
<td>General purpose – only below diamond for hardness</td>
<td>1150</td>
<td>550</td>
<td>3.4</td>
<td>0.85</td>
<td>1 – 12</td>
</tr>
<tr>
<td>Si</td>
<td>Excellent for far-IR spectral measurement</td>
<td>1150</td>
<td>8900-1500, 475-40</td>
<td>3.4</td>
<td>0.85</td>
<td>1 – 12</td>
</tr>
<tr>
<td>ZnSe</td>
<td>General purpose ATR crystal</td>
<td>120</td>
<td>520</td>
<td>2.4</td>
<td>2.00</td>
<td>5 – 9</td>
</tr>
</tbody>
</table>
# Attenuated Total Reflection (ATR)

## Chart of Common Crystal Materials

<table>
<thead>
<tr>
<th>Material</th>
<th>ATR Spectral Range (cm(^{-1}))</th>
<th>Refractive Index</th>
<th>Depth of Penetration ((\mu)) (at 45° &amp; 1000 cm(^{-1}))</th>
<th>Uses</th>
</tr>
</thead>
<tbody>
<tr>
<td>Germanium</td>
<td>5,500 - 675</td>
<td>4</td>
<td>0.66</td>
<td>Good for most samples. Strong absorbing samples, such as dark polymers.</td>
</tr>
<tr>
<td>Silicon</td>
<td>8,900 - 1,500 &amp; 360-120</td>
<td>3.4</td>
<td>0.85</td>
<td>Resistant to basic solutions.</td>
</tr>
<tr>
<td>AMTIR</td>
<td>11,000 - 725</td>
<td>2.5</td>
<td>1.77</td>
<td>Very resistant to acidic solutions.</td>
</tr>
<tr>
<td>ZnSe</td>
<td>15,000 - 650</td>
<td>2.4</td>
<td>2.01</td>
<td>General use.</td>
</tr>
<tr>
<td>Diamond</td>
<td>25,000 - 100</td>
<td>2.4</td>
<td>2.01</td>
<td>Good for most samples. Extremely caustic or hard samples.</td>
</tr>
</tbody>
</table>
Attenuated Total Reflection (ATR)

Materials of ATR crystals (IRE elements)

**Zinc Selenide ZnSe**
- preferred for all routine applications, limited use with strong acids and alkalies, surface etched during prolonged exposure to extremes of pH, complexing agents (ammonia and EDTA) will also erode its surface because of the formation of complexes with the zinc.

**AMTIR**
- as a glass from selenium, germanium and arsenic, insolubility in water, similar refractive index to zinc selenide, can be used in measurements that involve strong acids.
Attenuated Total Reflection (ATR)

Materials of ATR crystals (IRE elements)

Germanium Ge

- high refractive index, used when analyzing samples have a high refractive index

Silicon Si

- hard and brittle, chemically inert, affected only by strong oxidizers, well suited for applications requiring temperature changes as it withstands thermal shocks better than other ATR materials, hardest crystal material offered except for Diamond, which makes it well suited for abrasive samples that might otherwise scratch softer crystal materials, below 1500 cm$^{-1}$ usefulness limited
Attenuated Total Reflection (ATR)

Materials of ATR crystals (IRE elements)

**Diamond**

- for analysis of a wide range of samples, including acids, bases, and oxidizing agents, scratch and abrasion resistant, expensive, intrinsic absorption from approximately 2300 to 1800 cm\(^{-1}\) limits its usefulness in this region (5% transmission)
Attenuated Total Reflection (ATR)

For thin films, the ATR spectra are the same as transmission spectra.

For thick films, the absorption bands are more intense at longer wavelengths.

As the angle of incidence approaches the critical angle, the bands tend to broaden on the long wavelength side and the minima are displaced to longer wavelengths (lower wavenumbers). Dispersion type spectra are observed very close to and below critical angle.
Attenuated Total Reflection (ATR)

Experimental Setup - horizontal arrangement (HATR)

Single Bounce ATR
- Small sampling area
  - use for strong absorbers
  - solid samples, liquids

Multi-Bounce ATR
- Broad sampling area provides
  - greater contact with the sample
  - use for weak absorbers or dilute solutions
Attenuated Total Reflection (ATR)

Experimental Setup

- Sample clamp with micrometer controlled spring tension (other clamps also available)
- Mounting ring
- Teflon liquid holder
- Sampling area
- Liquid plate (trough) insert
- Universal ATR plate
- Clamp position control knob
- Base assembly

- Single Reflection ATR Plate
- 3 Reflection ATR Plate (includes 5mm swivel tip)
Attenuated Total Reflection (ATR)

MIRacle Pressure Clamps are Pinned-in-Place and Easily Upgraded

MIRacle Digital Clamp
Ideal for Controlled Pressure

MIRacle Rotating Clamp
Ideal for Cleaning Tip of Debris

MIRacle Viewing Clamp
Ideal for Placing Fibers or Crystals

MIRacle High-Pressure Clamp – Ideal for Routine Sampling

MIRacle Micrometer Clamp – OK for Low Pressure Applications

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**Slide 12**

Sample should completely cover the ZnSe Crystal indicated with the arrow below.

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**Slide 13**

Be sure that the press is rotated completely to the lowest level.
Attenuated Total Reflection (ATR)

Multi-Bounce ATR

HATR Accessory – in-compartment HATR for liquid and solid samples
Attenuated Total Reflection (ATR)

Scheme of the Circle ATR Cell
Attenuated Total Reflection (ATR)

Scheme of the Circle ATR Cell
Attenuated Total Reflection (ATR)

ATR Spectra

Thick Polymer Sample, using MIRacle with AMTIR Crystal and High-Pressure Clamp – No Sample Preparation.
Attenuated Total Reflection (ATR)

ATR Spectra

FTIR spectrum of cleansing lotion on human forearm – using the HATRPlus with flat plate Ge crystal.
Attenuated Total Reflection (ATR)

IR Spectra of Methanol

ATR (ZnSe 45°), ATR uncorrected

ATR (ZnSe 45°), corrected

Transmittion in the cell
Attenuated Total Reflection (ATR)

ATR Spectra of Rubber - Diamond vs Germanium IRE

- Need versatility: Diamond is excellent for most materials; Ge excels for HCB rubbers

Refractive indexes: Ge=4
Diamond=2.4
Summary:

- Versatile and *non-destructive* technique for variety of materials - soft solid materials, liquids, powders, gels, pastes, surface layers, polymer films, samples after evaporation of a solvent ..... 

- Requires minimal or *no sample preparation*

- Useful for *surface* characterization, opaque samples

Limitation: sensitivity is typically 3-4 orders of magnitude less than transmission
Specular vs Diffuse Reflection

Specular reflection is defined as light reflected from a smooth surface (such as a mirror, any irregularities in the surface are small compared to $\lambda$) at a definite angle, whereas diffuse reflection is produced by rough surfaces that tend to reflect light in all directions. There are far more occurrences of diffuse reflection than specular reflection in our everyday environment.

What kind of reflections account for the column of light reflected off the water?

What would we see on the water if it were perfectly flat, unmoving?
Specular vs Diffuse Reflection

Figure 3

Specular Reflection

Diffuse Reflection
Specular Reflection

= External Reflection Spectroscopy

- introduced in the 1960s, much wider use in the 1970s
- light is reflected from a smooth (mirror-like) sample at a definite angle to record its spectrum
- spectroscopic technique for films deposited on, or pressed against reflective surfaces

- if surface absorbs a wavelength of light \( \Rightarrow \) its relative intensity is decreased

Angle of incidence = Angle of reflection
Light Rays Through Thin Films

\[ I \rightarrow R_{01} \rightarrow T_{01}F^2R_{12}T_{10} \rightarrow \text{etc} \]

\[ \text{medium 0} \]

\[ \text{sample 1} \]

\[ \text{medium 2} \]

\[ F = \exp(-4pk_1d/l) \]

\[ T_{01}F \]

\[ T_{01}F^2R_{12} \]

\[ T_{01}F^3R_{12}R_{10} \]

\[ T_{01}FT_{12} \]

\[ \text{etc} \]
Specular Reflection

If the surface is smooth like a mirror:
- reflection and the incidence angles are equal
- reflected beam retains the polarization characteristics of the incidence beam

Thin layers: 0.5-20 μm ⇒ angle ~20-60° ⇒ spectra similar to transmission ones

Monomolecular layers: angle ~60-85° ⇒ spectra predominantly a function of the refractive index
⇒ derivative shape of the bands arising from superposition of extinction coefficient and dispersion of refractive index
Specular Reflection

Refractive index

Absorbance index

K. Yamamoto and H. Ishida, Vibrational Spectrosc., 8, 1 (1994)
Specular Reflection

Experimental Setup

- selection of incident angle

Incident angle influences
- effective pathlength
- polarized IR response
Specular Reflection

Experimental Setup
Specular Reflection

Specular Reflection Spectra

Absorbance index

Refractive index

K. Yamamoto and H. Ishida, Vibrational Spectrosc., 8, 1 (1994)
Specular Reflection

Correction of "Restrahlen" bands
Specular Reflection

Specular Reflection Spectra

FTIR spectrum of thiol monolayer measured using the VeeMAX II specular reflectance accessory set at 80 degrees angle of incidence, ZnSe polarizer and MCT detector.
Specular Reflection

Summary:

- non-destructive measurement of thin layers or monolayers
  - coatings on polished metals ⇒ e.g. varnishes
  - surface characterization

Limitation: Spectra depend on refractive index

- used for testing of lubricated surfaces of hard disks, to degradation studies of a protective coating on the surface; analysis of polymers on the surface of food containers and many others
Diffuse Reflection

Diffuse Reflectance Infrared Fourier Transform Spectroscopy (DRIFTs)

Spectra of powders and rough surfaces can be recorded by illuminating these surfaces and collecting sufficient scattered radiation with ellipsoids and paraboloids.

- fast measurement of powdered samples
- low repeatability of spectral data
- complicated physical description of the effect

shape of particles, compactness of samples
refractive index of particles
reflectivity and absorption characteristics of particles
Diffuse Reflection

Mechanisms generating infrared spectrum of a powder
Diffuse Reflection

Experimental Setup

Large and Small Sample Cups
Diffuse Reflection

Experimental Setup - Different Geometry
Diffuse Reflection

**DRIFTs Sample Preparation**

- powder **simply placed** into the sample cup and analyzed (no sample preparation)
- if the sample is too absorbent, it must be diluted in a nonabsorbent matrix (KBr, KCl,...)
  - no pellet pressing required
- particle size smaller than 10 \( \mu \text{m} \) (i.e. not exceeding the wavelength of the incident radiation) preferred
- or **paper disc** coated on a surface by SiC as an abrasive material \( \Rightarrow \) grinding of sample
- ideal for pharmaceutical and forensic applications
Diffuse Reflection

DRIFTs Spectra - y axe units

From the theoretical standpoint, there is no linear relation between band intensity and concentration (as valid in transmission), and quantitative analyses by the DRIFTS method are therefore rather complicated.

- expressed in linear units Kubelka-Munk (roughly correspond to absorbance in transmission KBr pellet technique)
Diffuse Reflection

DRIFTs Spectra - y axe units

- DRIFTs bands stronger than expected absorption from weak IR bands \( \Rightarrow \) compensation by Kubelka-Munk conversion

\[
f(R) = \frac{\left(1 - R_\infty \right)^2}{2R_\infty} = 2.303.a \cdot \frac{c}{s} = \frac{k}{s} \quad k = 2.303 \cdot a \cdot c
\]

- \( f(R) \) is called Kubelka-Munk function

\( R_\infty \) ... absolute reflectance of the sampled layer

\( k \) ... molar absorption coefficient \( a \) ... absorptivity

- proportional to the fraction of transmitted light

\( s \) ... diffusion (scattering) coefficient

- proportional to the fraction of diffused light

\( \Rightarrow \) creates a linear relationship for spectral intensity relative to sample concentration \( s \sim \text{const.} \Rightarrow f(R) = 2.303 \cdot a \cdot c \)
Diffuse Reflection

DRIFTS Spectra - y axle units

linear relationship assumed when:
- infinite sample dilution in a non-absorbing matrix, i.e. KCl, KBr, ... (such that $k = 0$ and $R = 1$)
- a constant scattering coefficient
- “infinitely thick” sample layer

These conditions achieved for:
- highly diluted samples
- small particle samples (the scattering coefficient is a function of sample size and packing)
- sample layer of at least 1.5 mm

With proper sample preparation DRIFTS can provide ppm sensitivity and high quality results.

A special version of the EasiDiff for NIR measurements (gold coated optics) is also available.
Diffuse Reflection

IR Spectra of CaCO$_3$

![Graph showing IR spectra of CaCO$_3$]
Diffuse Reflection

IR Spectra of 1,2-Bis(diphenyl phosphino)ethane

Diffuse Reflection vs Transmission

Zeolite Y14 (Si/Al of 13.6)

a) transmission spectrum, selfsupported pure powder pellet, after overnight activation at 723 K under vacuum

b) diffuse reflection spectrum after 10 h activation under nitrogen.

tspectra recorded at 298 K
Diffuse Reflection

Pt/Al$_2$O$_3$ DRIFTs spectra, variation in species at the surface during 1 h of nitration at 423 K (background first spectrum, activated surface).
Diffuse Reflection

DRIFTs spectra of Ca oxalate + KBr (5/95 w/w)
FT-IR Reflection Techniques

Summary:

Attenuated Total Reflection (ATR)
- structural information from the surface - thick layers

Specular Reflection (smooth surfaces)
- measurement of thin layers or monolayers
  (coatings on metals, surface characterization)

Diffuse Reflection (DRIFTs) (rough surfaces)
- structural information is from the bulk matrix