Potential energy atom-atom

London (dispersion) forces: longer separations, always attractive Fluctuating dipole – fluctuating dipole model:

- \bullet elst. field $E \propto 1/r^3$
- \bigcirc induced dipole $\mu_{ind} \propto E$
- energy $u(r) \propto \mu E \propto 1/r^6$ (negative = attractive)

Repulsion at shorter separations:

A, B, C are positive constants

$$u(r) \propto e^{-Br}$$

Total: exp-6

also Buckingham, Born-Mayer(-Huggins), Tosi-Fumi, . . . :

$$u(r) = Ae^{-Br} - \frac{C}{r^6}$$

Component of any atom-atom interaction

NB: " \propto " = "is proportional to"

Lennard-Jones potential

Repulsive forces approximated:

$$Ae^{-Br} \rightarrow \frac{A'}{r^{12}}$$

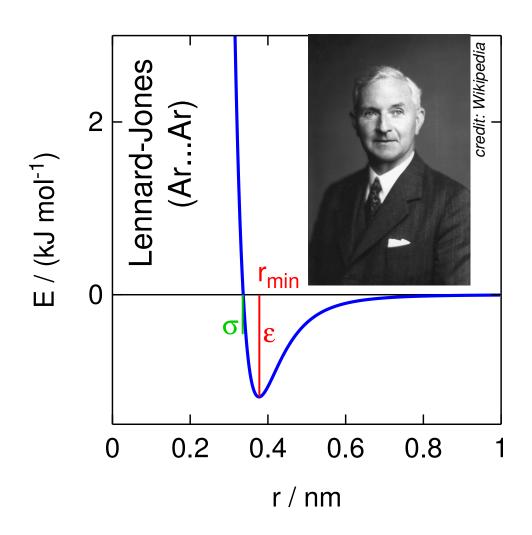
Common formula:

$$u(r) = 4\epsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^{6} \right]$$

$$E_{\min} = -\epsilon$$
, $r_{\min} = 2^{1/6}\sigma$

Optional formula:

$$u(r) = E_{\min} \left[2 \left(\frac{r_{\min}}{r} \right)^6 - \left(\frac{r_{\min}}{r} \right)^{12} \right]$$



Many molecules

...e.g., Ar

Approximation of **pair additivity**, accuracy ≈ 90 %

$$E_{\text{pot}} = \sum_{ij} u(r_{ij})$$

Better:

$$E_{\text{pot}} = \sum_{ij} u(r_{ij}) + \sum_{ijk} u_3(r_{ij}, r_{ik}, r_{jk})$$

where

$$u_3(r_{ij}, r_{ik}, r_{jk}) = u(\mathbf{r}_i, \mathbf{r}_j, \mathbf{r}_k) - u(r_{ij}) - u(r_{ik}) - u(r_{jk})$$

Electrostatic forces

charge-charge (ions)

$$U = \frac{1}{4\pi\epsilon_0} \frac{q_i q_j}{r_{ij}}$$

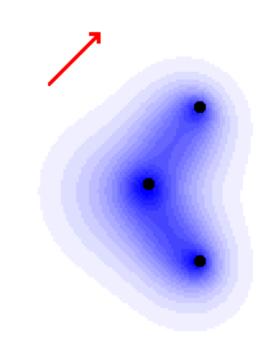
- partial charges: such charges placed at nuclei so that their electric field approximates well the real one
- dipole moment

$$\vec{\mu} = \sum_{i} q_{i} \vec{r}_{i}$$

polarizability (el. field induces a dipole)

$$\vec{\mu}_{\text{ind}} = \alpha \vec{E}$$

(is not pair additive)



Force field

Force field = PES as a sum of contributions, comprises their functional forms and tables of parameters

atomistic

Small molecules: rigid, rotations (water, CO, CH₄)

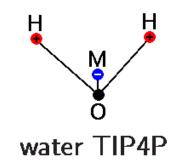
Large molecules: many terms

- bonded forces: vibrating bonds (1–2), angles (1–3) torsions (1–4) and dihedral potentials
- non-bonded forces (partially 1–4, 1–more): Lennard-Jones etc., charge–charge

2 3

Models:

- full-atom
- \bigcirc united-atom (-CH₃, -CH₂-, etc.)
- auxiliary interaction centers (TIP4P)
- coarse-grained



Bonded forces - bonds

Harmonic approximation:

$$U = K(r - r_0)^2$$

optionally:

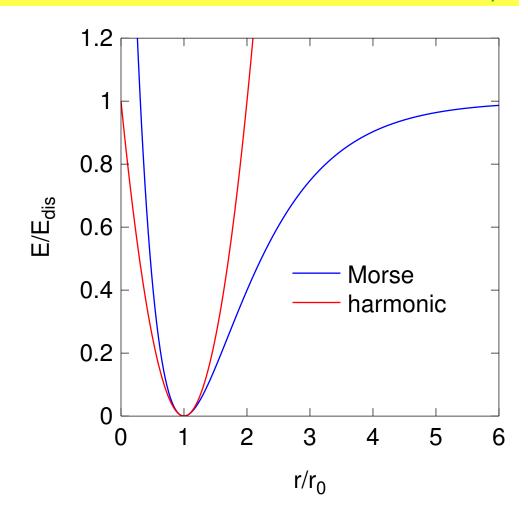
$$U = \frac{K'}{2}(r - r_0)^2$$

Fixed bond length:

$$r = r_0$$

Morse (dissociation):

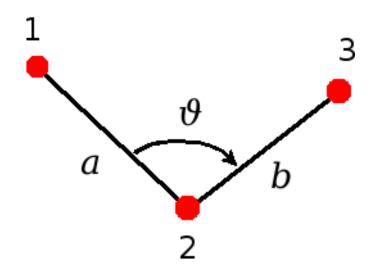
$$U = E_{dis} \left[1 - e^{-\alpha(r - r_0)} \right]^2$$



Bonded forces – angles

Harmonic approximation:

$$U(9) = K_{\text{harm}}(9 - 9_0)^2$$



Bonded forces - torsions

Dihedral potential (proper torsion)

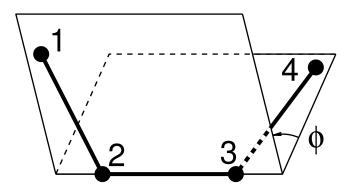
$$U(\phi) = \sum_{n} K_n \cos(n\phi)$$

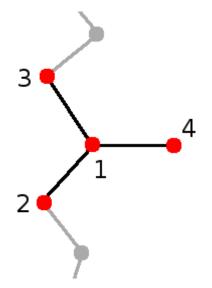
Non-bonded 1–4 terms scaled down by certain q (often q=0.5) are usually added. Note that the total dihedral potential is a sum of $U(\phi)$ and 1–4.

To keep the aromatic ring planar:

$$U(\phi) = \sum_{n} K_0 \phi^2$$

Improper torsion – keep >C=O etc. planar: the same form, different order of atoms





Non-bonded forces: combining rules

Lennard-Jones is defined by σ_i , ϵ_i . The energy of two identical atoms is

$$u_{ii}(r) = 4\epsilon_i \left[\left(\frac{\sigma_i}{r} \right)^{12} - \left(\frac{\sigma_i}{r} \right)^6 \right]$$

But what about two different atoms? (There are $\binom{N}{2}$ pairs!).

Lorentz-Berthelot combining rule (geometric mean for energy, arithmetic mean for diameters, better for vapor-liquid equilibria):

$$\epsilon_{ij} = \sqrt{\epsilon_i \epsilon_j}, \quad \sigma_{ij} = \frac{\sigma_i + \sigma_j}{2}$$

Geometric rule (better for crystals):

$$\epsilon_{ij} = \sqrt{\epsilon_i \epsilon_j}, \quad \sigma_{ij} = \sqrt{\sigma_i \sigma_j}$$

... more exist.

Force field development

- geometry: spectroscopy, diffraction, quantum chemistry calculations (ab initio, DFT)
- bonded forces: quantum chemistry calculations, spectroscopy
- igoplus Lennard-Jones σ : experimental density, structure
- \bigcirc Lennard-Jones ϵ : vaporization enthalpy

$$\langle U_{\text{pot, intermol.}} \rangle = \Delta_{\text{vap}} U \approx \Delta_{\text{vap}} H - nRT$$

- repulsions more precisely: compressibility, elastic moduli in crystals
- partial charges:
 - dipole moments: spectroscopy, permittivity (dielectric constant)
 - quantum chemistry calculations (Mulliken, CHELPG = CHarges from Electrostatic Potentials using a Grid based method)
- and/or clusters (by quantum chemistry)
- opolarizability: experiment, quantum chemistry calculations
- fine-tuning (of partial charges etc.): diffusivity
- structure (radial distribution functions); reverse MC

External forces

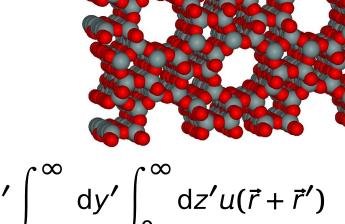
Electrostatic, gravitational ...

Walls, pores:

- made of atoms
- hard wall

$$U_{\text{hard wall}}(\vec{r}) = \begin{cases} \infty, & \text{for } z < 0, \\ 0 & \text{for } z \ge 0 \end{cases}$$

igcup integrated (soft) wall of number density* $\mathcal{N} = N/V$



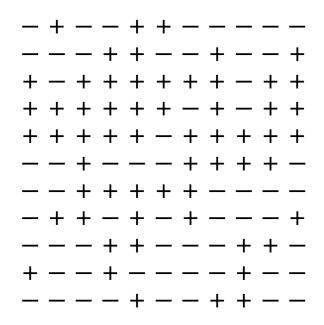
$$U_{\text{soft wall}}(\vec{r}) = \mathcal{N} \int_{z'>0} u(\vec{r} + \vec{r}') d\vec{r}' = \mathcal{N} \int_{-\infty}^{\infty} dx' \int_{-\infty}^{\infty} dy' \int_{0}^{\infty} dz' u(\vec{r} + \vec{r}')$$

Lennard-Jones → 3-9 potential

$$U_{\text{LJ-wall}}(\vec{r}) = -2\pi\epsilon\mathcal{N}\sigma^{3} \left[\frac{1}{3} \left(\frac{\sigma}{z} \right)^{3} - \frac{2}{45} \left(\frac{\sigma}{z} \right)^{9} \right]$$

^{*}also denoted ρ or n

Lattice models: Ising



Ferromagnet:

$$U = -J \sum_{\langle i,j \rangle} s_i s_j + h \sum_i s_i,$$

$$s_i \in \{-1, +1\} = \{\downarrow, \uparrow\}$$

/ = interaction constant:

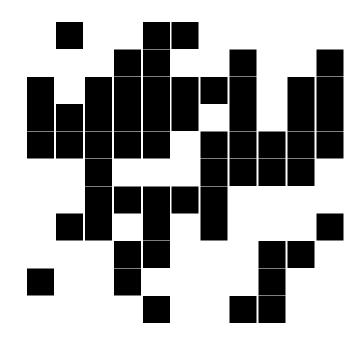
J > 0: ferromagnet,

J < 0: antiferromagnet

h = magnetic field intensity

Critical (Curie) point: $h_C = 0$;

2D: $T_c/J = 2/\ln(1+\sqrt{2})$



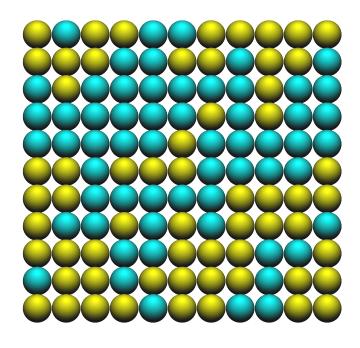
Lattice gas:

$$U = -\epsilon \sum_{\langle i,j \rangle} n_i n_j + \mu \sum_i n_i,$$

$$n_i \in \{0, 1\} = \{ , \}$$

 ϵ = attraction constant μ = chemical potential Equivalence:

$$n_i = (1 + s_i)/2$$

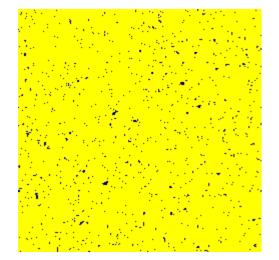


Binary alloy:

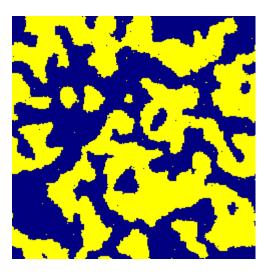
$$U = -\sum_{\langle i,j \rangle} \epsilon_{k_i k_j} + \sum_i \mu_{k_i},$$
$$k_i \in \{ \bigcirc, \bigcirc \}$$

 $\epsilon_{\bullet,\bullet}$, $\epsilon_{\bullet,\bullet}$, $\epsilon_{\bullet,\bullet}$ = nearest-neighbor interactions μ_{\bullet} , μ_{\bullet} = chem. potentials Equiv.: $n_i = 0 \sim k_i = \bullet$

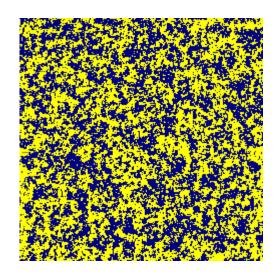
$$n_i = 1 \sim k_i = \bigcirc.$$



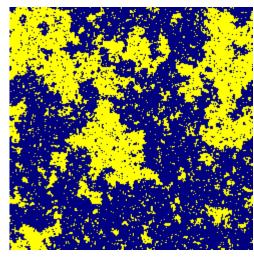
low temperature $0.8T_{crit}$



quenched system $5T_{crit} \rightarrow 0.5T_{crit}$



high temperature $1.25T_{krit}$



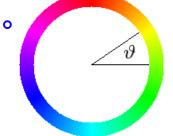
critical point T_{crit}

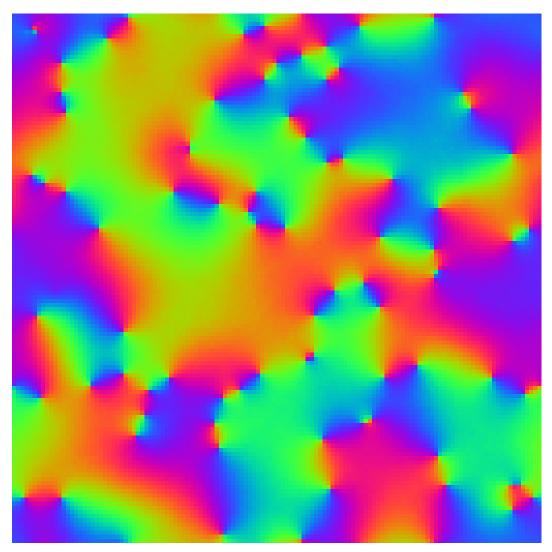
Lattice models: XY

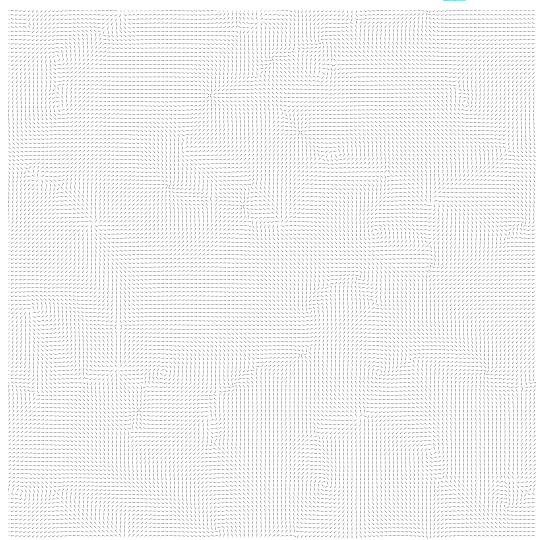
 $+\frac{14/15}{502/2}$

Site $i \mapsto$ continuous 2D "spin" $\theta_i \in [0, 2\pi) = 0^\circ$, $= 120^\circ$, $= 240^\circ$

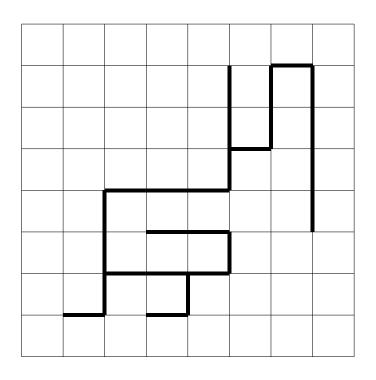
$$\mathcal{H} = -J \sum_{\langle i,j \rangle} \cos(\theta_i - \theta_j) + h \sum_i \cos(\theta_i)$$

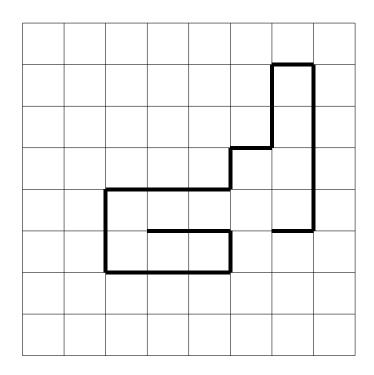






Lattice model of polymer





no branching = self-avoiding random walk