

Problem 15-03 Spreading of a liquid on solid surface

A small amount of liquid was placed on a plane surface of a solid, the surface energy of which was 48.5 mJ m^{-2} . How will it behave? Will it spread? The interfacial tension liquid-solid surface has the value of 18.3 mN m^{-1} . The surface tension of the liquid has been determined by the capillary rise method: in the vertical capillary of inner diameter 0.16 mm the elevation of the liquid of 5.7 cm was measured. The liquid density is 0.912 g cm^{-3} and the studied liquid completely wets the walls of the capillary.

$$[\gamma_{\text{sg}} > \gamma_{\text{sl}} + \gamma_{\text{lg}} ; S_{\ell/\text{s}} = 9.8 > 0, \cos \theta > 1 - \text{liquid spreads} (\gamma_{\text{lg}} = 20.4 \text{ mN m}^{-1})]$$

Solution:

$$\gamma_{\text{sg}} = 48.5 \text{ mJ m}^{-2}$$

$$\gamma_{\text{sl}} = 18.3 \text{ mN m}^{-1}$$

γ_{lg} –by capillary method:

$$\begin{aligned} D &= 0.16 \text{ mm} = 1.6 \cdot 10^{-4} \text{ m} \\ R &= 8 \cdot 10^{-5} \text{ m} \\ h &= 5.7 \text{ cm} = 0.057 \text{ m} \\ \rho &= 0.912 \text{ g cm}^{-3} = 912 \text{ kg m}^{-3} \end{aligned}$$

$$\pi R^2 \cdot h \cdot \rho \cdot g = 2 \pi R \cdot \gamma_{\text{lg}}$$

$$\gamma_{\text{lg}} = \frac{R \cdot h \cdot \rho \cdot g}{2} = \frac{8 \cdot 10^{-5} \cdot 0.057 \cdot 912 \cdot 9.81}{2} = 0.0204 \text{ N m}^{-1} = 20.4 \text{ mN m}^{-1}$$

Spreading condition:

- Harkins spreading coefficient

$$\begin{aligned} S_{\ell/\text{s}} &= \gamma_{\text{sg}} - \gamma_{\text{sl}} - \gamma_{\text{lg}} = \\ &= 48.5 - 18.3 - 20.4 = +9.8 > 0 - \text{liquid spreads} \end{aligned}$$

or

- Young equation:

$$\cos \theta = \frac{\gamma_{\text{sg}} - \gamma_{\text{sl}}}{\gamma_{\text{lg}}} = \frac{48.5 - 18.3}{20.4} = 1.48 > 1$$

- $\cos \theta > 1$ is impossible, the drop cannot take an equilibrium shape and the liquid spreads

