

Mechanical quantities

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- Temperature (NVE MD):

$$T_{kin} = \frac{E_{kin}}{fk_B/2}$$
 notation here:
 $U = U(r^N)$ = potential energy
 $E = E(T, V)$ = internal energy
 f = # of degrees of freedom
- Internal energy:

$$E = (E_{kin} + U) \stackrel{NVT}{=} \frac{f}{2} k_B T + \langle U \rangle \equiv E_{id} + E_{res}$$
 res = residual
 see next slide
- Pressure

$$p = \frac{N}{V} k_B T - \left\langle \left(\frac{\partial U(V^{1/3} \xi^N)}{\partial V} \right)_{\xi^N} \right\rangle \equiv p_{id} + p_{res}$$

$$\beta = 1/k_B T$$
 - dimensionless (scaled) coordinates $\xi_i: r_i = V^{1/3} \xi_i$
 - red derivative is calculated at constant ξ^N
 - whole configurations is uniformly shrank/swelled
 - p_{id} = kinetic contribution (=ideal gas), also $p_{id} = \phi 2E_{kin}/3V$, where in periodic b.c. $\phi = N/(N-1)$ takes into account 3 zero conserved momenta
 - p_{res} = cohesion contribution

Pressure tensor

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Also called stress tensor (in solids):

$$\vec{P} = \vec{P}_{id} + \vec{P}_{res} = \frac{1}{V} \sum_{i=1}^N (\phi m_i \vec{v}_i \vec{v}_i + \vec{f}_i \vec{r}_i)$$

- Tensor product $\vec{T} = \vec{u} \vec{v}$, also denoted $\vec{T} = \vec{u} \otimes \vec{v}$; $T_{ab} = v_a v_b$.
- $\phi = N/(N-1)$ takes into account 3 zero conserved momenta in periodic b.c.
- The scalar pressure is 1/3 of its trace, $p = \text{tr}(\vec{P})/3$.
- In models with rigid (constrained) bonds, the virial of constrained forces has to be included (it depends on velocities).
- Diagonal terms are good for calculating surface tension.
- Off-diagonal terms are good for calculating viscosity.*
- Can be also calculated by the virtual box change:
 - elongation in $\hat{x} \Rightarrow P_{xx}$,
 - simulation cell shape change \Rightarrow off-diagonal terms.

*Surprisingly, the diagonal terms can be used to calculate viscosity, too.

Pressure in the NVT ensemble: Derivation

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$$dF = -SdT - pdV \quad (X) = \frac{1}{Q_N} \int_{V^N} X(r^N) \exp[-\beta U(r^N)] d^N r^N$$

$$p = - \left(\frac{\partial F}{\partial V} \right)_T, \quad F = -k_B T \ln \frac{Q_N}{N! \Lambda^{3N}}$$

$$Q_N = \int_{V^N} \exp[-\beta U(r^N)] d^N r^N \quad r_i = V^{1/3} \xi_i \quad \int_{1^N} \exp[-\beta U(V^{1/3} \xi^N)] V^N d\xi^N$$

$$p = - \left(\frac{\partial F}{\partial V} \right)_T = k_B T \left(\frac{\partial \ln Q_N}{\partial V} \right)_{\xi^N} = \frac{k_B T}{Q_N} \left(\frac{\partial Q_N}{\partial V} \right)_{\xi^N}$$

$$= \frac{k_B T}{Q_N} \int_{1^N} \exp[-\beta U(V^{1/3} \xi^N)] N V^{N-1} d\xi^N$$

$$+ \frac{k_B T}{Q_N} \int_{1^N} \exp[-\beta U(V^{1/3} \xi^N)] (-\beta) \left(\frac{\partial U(V^{1/3} \xi^N)}{\partial V} \right)_{\xi^N} V^N d\xi^N$$

$$= \frac{N}{V} k_B T - \left\langle \left(\frac{\partial U(V^{1/3} \xi^N)}{\partial V} \right)_{\xi^N} \right\rangle = \text{ideal part (kinetic)} + \text{residual part (correction)}$$

Surface tension of liquid

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Using slab geometry – elongated periodic box (2:1 or more)

$$\gamma = \left(\frac{\partial F}{\partial A} \right)_{V,T} = \left(\frac{\partial G}{\partial A} \right)_{p,T} = -\frac{1}{4} L_z P_t, \quad \text{where } P_t = P_{xx} + P_{yy} - 2P_{zz}$$

- It is “mechanical quantity”.
- Cannot be used for interfacial energy of crystals (it is “entropic quantity”).
- Cutoff corrections – several variants (cf. simenon9).
- P_{zz} = saturated vapor pressure; if small enough, P_{xx}, P_{yy} can be replaced by the usual pressure p .
- Alternative: virtual area change method [Gloor et al.: JCP 123, 134703 (2005)]: scaling in \hat{x}, \hat{y} and inversely squared in \hat{z} so that volume is preserved:

$$\gamma = \left\langle \left(\frac{\delta U}{\delta A} \right)_V \right\rangle$$

Residual quantities

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= with respect to the standard state of ideal gas at the same temperature, volume (= density), and composition as the given system. Useful in the canonical (NVT) ensemble.

sometimes “excess”; for solvation Gibbs energy or μ of solute, also “Ben-Naim standard state”

For the Helmholtz energy:

$$F = -k_B T \ln Z_N = -k_B T \ln \frac{Q_N}{N! \Lambda^{3N}} = -k_B T \ln \frac{V^N}{N! \Lambda^{3N}} - k_B T \ln \frac{Q_N}{V^N} \equiv F_{id} + F_{res}$$
 ideal gas: $Q_N = V^N$

Refresh:
 Thermal de Broglie wavelength:

$$\Lambda = \frac{h}{\sqrt{2\pi m k_B T}}$$

chemical potential of ideal gas:

$$\mu_{id} = \left(\frac{\partial F_{id}}{\partial N} \right)_{T,V} = k_B T \ln \frac{N \Lambda^3}{V}$$

Fluctuation quantities

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$\langle (\Delta X)^2 \rangle = \text{Var } X = (\text{mean quadratic fluctuation} = \text{variance})$
 $\langle \Delta X \rangle = X - \langle X \rangle$
 fluctuation = (mech/el/... quantity)' = (thermodynamic potential)''

Less accurate than mean values!
 E.g. (NVT): $P = - \left(\frac{\partial F}{\partial V} \right)_T, E = \left(\frac{\partial F}{\partial \beta} \right)_V$, then:

- Heat capacity at constant [V]:

$$C_V = \left(\frac{\partial E}{\partial T} \right)_V = \frac{1}{k_B T^2} \text{Var } E = \frac{1}{k_B T^2} \langle (\Delta E_{kin} + \Delta U)^2 \rangle$$

$$\text{Cov}(U, E_{kin}) = \Delta U \Delta E_{kin} = 0, \text{Var } E_{kin} = \frac{1}{2} (k_B T)^2 \text{ (see exercise)} \Rightarrow$$

$$C_V = \frac{1}{k_B T^2} \left\langle \left(\frac{f k_B T}{2} + \Delta U \right)^2 \right\rangle = \frac{f k_B}{2} + \frac{1}{k_B T^2} \langle (\Delta U)^2 \rangle \equiv C_{V,id} + C_{V,res}$$
- Isothermal compressibility – in the NPT ensemble

$$\kappa = -\frac{1}{V} \left(\frac{\partial V}{\partial P} \right)_T = \frac{\langle (\Delta V)^2 \rangle}{V k_B T}$$

Permittivity can be calculated from the fluctuation of the cell dipole moment, $\langle M^2 \rangle$, (more later).

Pressure – virtual volume change method

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$$p = \frac{N}{V} k_B T - \left\langle \left(\frac{\partial U(V^{1/3} \xi^N)}{\partial V} \right)_{\xi^N} \right\rangle$$

Numerical derivative (for a selected series of configurations)

$$\frac{\partial U}{\partial V} = \frac{U(V + \Delta V) - U(V)}{\Delta V} + \mathcal{O}(\Delta V) \equiv \frac{U(\xi^N, V + \Delta V) - U(\xi^N, V)}{\Delta V} + \mathcal{O}(\Delta V)$$

$$\frac{\partial U}{\partial V} = \frac{U(V + \Delta V) - U(V - \Delta V)}{2\Delta V} + \mathcal{O}(\Delta V^2)$$

Implementation: $U(V + \Delta V)$ means that the whole configuration (all distances) is swelled by the same ratio; for molecules w.r.t. a reference point (then, N = # of molecules):

$$\left(\frac{V + \Delta V}{V} \right)^{1/3}$$

The scaled configuration is not included in the trajectory.

For models with a hard core such that swelling the box cannot cause an overlap, shrinking can be used: $P = N k_B T / V + \frac{k_B T}{\Delta V} (e^{-[U(V-\Delta V) - U(V)]/k_B T} + \mathcal{O}(\Delta V))$

Exercise

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Calculate $\langle E_{kin} \rangle$ and $\text{Var}(E_{kin})$ for one degree of freedom, $E_{kin} = \frac{1}{2} m v^2$.

$$E_{kin} = \frac{1}{2} m v^2$$

$$\pi(v) = \frac{\exp(-\frac{1}{2} m v^2 / k_B T)}{\int \exp(-\frac{1}{2} m v^2 / k_B T) dv}$$

$$\langle E_{kin} \rangle = \int E_{kin} \pi(v) dv$$

$$\text{Var}(E_{kin}) = \int (E_{kin} - \langle E_{kin} \rangle)^2 \pi(v) dv = \frac{1}{2} (k_B T)^2$$

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> restart;
> assume(m>0, k>0, T>0);
> k:=m/2*v^2: KT:=k*T:
> inorm:=int(exp(-K/KT),v=-infinity..infinity);
   inorm:=sqrt(k*T)/sqrt(m);
> averk:=int((K*exp(-K/KT),v=-infinity..infinity)/inorm;
   averk:=1/2*k*T:
> fluctk:=int((K-averk)^2*exp(-K/KT),v=-infinity..infinity)/inorm;
   fluctk:=1/2*k*T^2:

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Pressure from the virial of force

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The derivative expanded:

$$\frac{\partial U(V^{1/3} \xi^N)}{\partial V} = \sum_{i=1}^N \frac{1}{3} V^{-2/3} \xi_i \cdot \frac{\partial U}{\partial r_i} = \frac{1}{3V} \sum_{i=1}^N \vec{r}_i \cdot \frac{\partial U}{\partial \vec{r}_i}$$

The result is

$$p = \frac{N}{V} k_B T + \frac{1}{3V} \langle W_f \rangle, \quad W_f = - \sum_{i=1}^N \vec{r}_i \cdot \frac{\partial U}{\partial \vec{r}_i} = \sum_{i=1}^N \vec{r}_i \cdot \vec{f}_i \text{ (virial of force)}$$

... cannot be directly applied in the periodic boundary conditions.

- Pair additivity in the periodic boundary conditions \Rightarrow

$$p = \frac{N}{V} k_B T - \frac{1}{3V} \sum_{\langle ij \rangle} \langle r_{ij} u'(r_{ij}) \rangle \equiv p_{id} + p_{res}$$
- For molecular models one can use either the atom-frame (N = # of atoms/sites), or molecular frame (N = # of molecules; typically, ref. point = center of mass). Of course, the formulas differ.

Isochoric heat capacity C_V in the NVE ensemble

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In the **canonical** ensemble:

$$\text{Var } X = k_B T^2 \frac{\partial^2 X}{\partial T^2} \equiv k_B T^2 X' \quad (X = U, E_{kin}) \quad (1)$$

$$\text{Cov}(U, E_{kin}) = 0$$

$$C_V = U' + E'_{kin}, \quad E'_{kin} = \frac{f k_B}{2}$$

Unnormalized probability distribution for temperature T :

$$w(U, E_{kin}) \sim \exp \left[-\frac{(\delta U)^2}{2 \text{Var } U} - \frac{(\delta E_{kin})^2}{2 \text{Var } E_{kin}} \right]$$

where the deviations are linearized: $(U, E_{kin}) = (U_0 + \delta U, E_{kin,0} + \delta E_{kin})$

Unnormalized probability distribution taking into account change in temperature, $T = T_0 + \delta T$:

$$w(T, U, E_{kin}) \sim \exp \left[-\frac{(\delta U - U' \delta T)^2}{2 \text{Var } U} - \frac{(\delta E_{kin} - E'_{kin} \delta T)^2}{2 \text{Var } E_{kin}} \right]$$

Isochoric heat capacity C_V in the NVE ensemble II

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In the **MD NVE** ensemble it holds $U + E_{kin} = E = \text{const}$ or $U = E - E_{kin}$, and of course $\delta E_{kin} = -\delta U$, so that let us choose δE_{kin} as the independent variable

$$w(T, E_{kin}) \sim \exp \left[- \left(\frac{1}{2k_B T^2 U'} + \frac{1}{2k_B T^2 E'_{kin}} \right) \delta E_{kin}^2 - \left(\frac{U'}{2k_B T^2} + \frac{E'_{kin}}{2k_B T^2} \right) \delta T^2 \right]$$

where the variances were replaced by derivatives, see (1).

NB: term $\delta E_{kin} \delta T$ cancels out!

From term at δT^2 :

$$\text{Var } T = \frac{k_B T^2}{U' + E'_{kin}} = \frac{k_B T^2}{C_V}$$

which is a known expression for the temperature in the NVE ensemble, however, T here is **not** the kinetic temperature, but T in formula $dE = T ds [V]$. The average temperature is the same, but not the fluctuation.

Examples

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For $\lambda = \beta$ we get as before:

$$\beta_1 F_{res}(\beta_1) - \beta_0 F_{res}(\beta_0) = \int_{\beta_0}^{\beta_1} \langle U \rangle d\beta$$

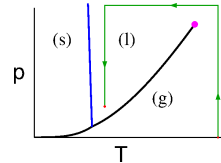
Integration from an Einstein crystal to a real crystal.

NB: Einstein crystal = independent (here classical) harmonic oscillators at lattice sites. There are minor solvable problems when the crystal is detached from the fixed sites.

Integration from ideal gas (e.g., around the critical point to liquid). Because of gas singularity at zero density, one of the integrals is (for NPT):

$$\ln \phi = \frac{\mu - \mu^o}{RT} = \int_0^p \left(V_m - \frac{RT}{p'} \right) dp'$$

It helps to use a virial equation of state at low densities.



Isochoric heat capacity C_V in the NVE ensemble III

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From the term at δE^2 :

$$\text{Var } E = \text{Var } U = \text{Var } E_{kin} = \frac{k_B T^2}{1/U' + 1/E'_{kin}} = \frac{k_B T^2}{1/(C_V - E'_{kin}) + 1/E'_{kin}}$$

Since E'_{kin} is known:

$$C_V = \frac{fk}{2} \left[\left(\frac{2T^2}{f \text{Var } T_{kin}} - 1 \right)^{-1} + 1 \right]$$

where $T_{kin} = E_{kin}/(fk/2)$ is the kinetic temperature.

Final expression applicable in simulations:

$$C_V = \frac{fk}{2} \left[\left(\frac{2 \langle T_{kin}^2 \rangle}{f \langle (\Delta T_{kin})^2 \rangle} - 1 \right)^{-1} + 1 \right]$$

Non-Boltzmann sampling

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We want $\langle BU \rangle_1$, but we simulate $\langle BU \rangle_0$ (can change β/U both)!

$$\Delta \langle BU \rangle = \langle BU \rangle_1 - \langle BU \rangle_0$$

$$\langle X \rangle_{(BU)_1} = \frac{\int X e^{-\beta U_1} dP^N}{\int e^{-\beta U_1} dP^N} = \frac{\frac{1}{Q_0} \int X e^{-\beta U_0} e^{-\Delta \beta U} dP^N}{\frac{1}{Q_0} \int e^{-\beta U_0} e^{-\Delta \beta U} dP^N} = \frac{\langle X e^{-\Delta \beta U} \rangle_0}{\langle e^{-\Delta \beta U} \rangle_0}$$

Helmholtz energy:

$$\begin{aligned} \Delta \langle BF_{res} \rangle &= \beta_1 F_{res}(\langle BU \rangle_1) - \beta_0 F_{res}(\langle BU \rangle_0) \\ &= -\ln \left(\frac{Q_1}{Q_0} \right) = -\ln \frac{\int e^{-\beta U_1} dP^N}{\int e^{-\beta U_0} dP^N} \\ &= -\ln \frac{\int e^{-\beta U_0} e^{-\Delta \beta U} dP^N}{\int e^{-\beta U_0} dP^N} = -\ln \langle e^{-\Delta \beta U} \rangle_0 \\ &= \ln \langle e^{+\Delta \beta U} \rangle_1 \end{aligned}$$

where the last equation follows from $0 \leftrightarrow 1$ interchange

First usage probably Zwanzig (1954) "thermodynamic perturbation method", loosely also "umbrella sampling"

Entropic quantities

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These include F, G (\leftarrow partition function), S (number of states W), $\mu, \Delta G, \dots$

They cannot be expressed as a simple $\langle \cdot \rangle$.

Usage: equilibria in general; solubility, ligand-receptor binding, stability of biomolecules, ...

thermodynamic integration:

- over a real variable (T, V, P)
- over a coupling parameter

reversible work calculated by the integration of force

Widom particle insertion method; gradual insertion, alchemical transmutation

non-Boltzmann sampling:

- umbrella sampling
- multiple histogram reweighting
- metadynamics/Wang-Landau/conformational flooding
- local density method

Formally (for the configurational integral):

$$Q = \frac{\int e^{-\beta U} dP^N}{1} = \frac{\int e^{-\beta U} dP^N}{V^{-N} \int e^{-\beta U} e^{+\beta U} dP^N} = \frac{V^N}{\langle e^{\beta U} \rangle}$$

... totally useless

Non-Boltzmann sampling contd.

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$\Delta \langle BU \rangle$ must not be too large

$$e^x = 1 + x + \dots$$

$$\ln(1+x) = x + \dots$$

the thermodynamic integration is recovered for infinitesimally small $\Delta \langle BU \rangle$:

$$\begin{aligned} \Delta \langle BF_{res} \rangle &= -\ln \langle e^{-\Delta \beta U} \rangle_0 \\ &\approx -\ln(1 - \Delta \langle \beta U \rangle_0) \\ &= -\ln(1 - \langle \Delta \beta U \rangle_0) \\ &\approx \langle \Delta \beta U \rangle_0 \end{aligned}$$

$$\text{general } \lambda: \frac{\partial \langle BF_{res} \rangle}{\partial \lambda} = \langle \partial \Delta \langle \beta U \rangle / \partial \lambda \rangle_\lambda$$

$$\lambda = \beta: \frac{\partial \langle BF_{res} \rangle}{\partial \beta} = \langle U \rangle$$

Umbrella sampling

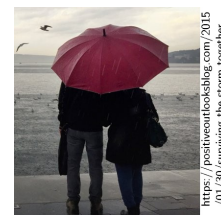
The system in the middle is sampled: $\text{mid} = (\beta U_0) + \Delta \langle \beta U \rangle / 2 = (\beta_0 U_0 + \beta_1 U_1) / 2$:

$$\Delta \langle BF_{res} \rangle = \ln \langle e^{+\Delta \beta U / 2} \rangle_{\text{mid}} - \ln \langle e^{-\Delta \beta U / 2} \rangle_{\text{mid}}$$

Nowadays, term "umbrella sampling" usually refers to many steps of similar kind.

Scaled particle

A particle growing step-wise, similar is "alchemical transmutation"



https://positivestorm.com/2015/01/30/surviving-the-storm-together

Thermodynamic integration

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Physical chemistry: $dF = -SdT - pdV, dG = -SdT + Vdp$

$U = U(r^N)$ = potential energy
 $E = E(T, V)$ = internal energy

Canonical ensemble ($\beta = 1/k_B T$):

$$\left(\frac{\partial F}{\partial V} \right)_T = -p \Rightarrow F_1 - F_0 = - \int_{V_0}^{V_1} p dV, \quad \left(\frac{\partial \langle BF \rangle}{\partial \beta} \right)_V = E \text{ or } \left(\frac{\partial \langle BF_{res} \rangle}{\partial \beta} \right)_V = \langle U \rangle$$

Numerically integrated: p, E must be determined at many points

Start from a suitable reference state (known state, ideal gas, harmonic crystal)

Proof # 1 of $\frac{\partial \langle BF \rangle}{\partial \beta} = E$:

$$\frac{\partial \langle BF \rangle}{\partial \beta} = \frac{\partial \langle F \rangle}{\partial \beta} = \frac{\partial \langle F \rangle}{\partial (1/T)} = \frac{\partial \langle F \rangle}{\partial T} \frac{\partial (1/T)}{\partial T} = \frac{-ST - F}{T^2} \cdot \left(-\frac{1}{T^2} \right) = ST + F = E$$

Proof # 2 of $\frac{\partial \langle BF \rangle}{\partial \beta} = E$:

$$\frac{\partial \langle BF \rangle}{\partial \beta} = -\frac{\partial \ln Z}{\partial \beta} = -\frac{1}{Z} \frac{\partial Z}{\partial \beta} = -\frac{\partial \sum_{\psi} e^{-\beta \mathcal{E}(\psi)}}{\partial \beta} = -\frac{\sum_{\psi} [-\mathcal{E}(\psi)] e^{-\beta \mathcal{E}(\psi)}}{\sum_{\psi} e^{-\beta \mathcal{E}(\psi)}} = \langle \mathcal{E} \rangle = E$$

Integration over a coupling parameter

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Let us consider any dependence $\langle BU \rangle(\lambda)$, e.g.:

$$\langle BU \rangle(\lambda) = \begin{cases} \beta[U_0 + \lambda(U_1 - U_0)] & \lambda = \text{coupling parameter, } \lambda \in [0, 1] \\ \lambda U & \lambda \equiv \beta: \text{ see previous slide} \end{cases}$$

then

$$\frac{\partial \langle BF_{res} \rangle}{\partial \lambda} = -\frac{\partial \ln Q}{\partial \lambda} = -\frac{1}{Q} \int \frac{\partial e^{-\beta U}}{\partial \lambda} dP^N = \frac{1}{Q} \int \frac{\partial \langle BU \rangle(\lambda)}{\partial \lambda} e^{-\beta U(\lambda)} dP^N = \left\langle \frac{\partial \langle BU \rangle(\lambda)}{\partial \lambda} \right\rangle_\lambda$$

$$\langle BF_{res} \rangle(\lambda_1) = \langle BF_{res} \rangle(\lambda_0) + \int_{\lambda_0}^{\lambda_1} \left\langle \frac{\partial \langle BU \rangle(\lambda)}{\partial \lambda} \right\rangle_\lambda d\lambda$$

where $\langle \cdot \rangle_\lambda$ = mean value in the ensemble (simulation) with potential $U(\lambda)$

How to integrate: (over a real variable or coupling parameter)

Several discrete values of λ_i :

- fit to a suitable function and integrate it
- use a numerical quadrature method; e.g., the Simpson formula (points with a higher weight should be simulated longer)

A little change of λ in every MD step (the system is almost in equilibrium) + integration

Multiple histogram reweighting I

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aka WHAM, weighted histogram analysis method

Building the **density of states** as a function of energy in a wide range of temperatures from overlapping histograms of energies obtained in a number of simulations at different temperatures (can be extended to other coupling parameter).

Configurational integral and residual Helmholtz energy at temperature $T_i, \beta_i = 1/k_B T_i$:

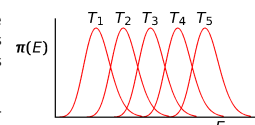
$$Q_i = e^{-\beta_i F_i} = \int e^{-\beta_i U(r^N)} dP^N = \int D(E) e^{-\beta_i E} dE$$

where $D(E)$ (aka W, Ω) is the density of states:

$$D(E) = \int_{U(r^N)=E} 1 dP^N \equiv \int \delta(U(r^N) - E) dP^N$$

$$\approx \int_{U(r^N) \in [E - \Delta E/2, E + \Delta E/2]} 1 dP^N$$

[Ghouri et al. (2008)]



$$\pi(E) = D(E) e^{-\beta E}$$

$$D(E) \propto E^{\text{const } N}$$

Detour: Density of states for a particle in a box

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One particle in 1D in a box of length L , energies of eigenstates according to the Schrödinger equation:

$$E_n = \frac{n^2 h^2}{8L^2} \propto n^2, \quad n = 1, 2, \dots$$

Let's have $f = 3N$ such degrees of freedom. The number of states $\#(E)$ with energy less than E satisfies the equation

$$n_1^2 + n_2^2 + \dots + n_f^2 < E = [E^{1/2}]^2$$

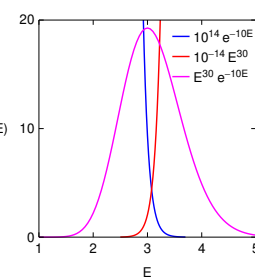
For large $E, \#(E) = 1/2^f$ of the volume of the $E^{1/2}$ -ball in the f -dimensional space:

$$\#(E) = \frac{\pi^{f/2} E^{f/2}}{\Gamma(f/2 + 1)} \propto E^{f/2} \Rightarrow D(E) = \frac{d\#(E)}{dE} \propto E^{f/2 - 1}$$

$$\pi(E) = D(E) e^{-\beta E}$$

The Boltzmann factor $e^{-\beta E}$ eventually wins!

For large N , the product converges to a Gaussian with $\text{Var}(E) \propto f \propto N$ ($\sigma = N^{1/2}$)



Multiple histogram reweighting II

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In the simulation, we calculate the histogram $h_i(E)$ for a set of (equidistant) energies E , or some equivalent Gaussian-based δ -function approximation. We will use the normalized histogram and the \int -form. (To repeat, subscript i refers to T_i .)

$$\sum_E h_i(E) = \int h_i(E) dE = 1$$

$$h_i(E) = \frac{D(E)e^{-\beta_i E}}{\int D(E)e^{-\beta_i E} dE} = D(E)e^{-\beta_i(E-F_i)}$$

Using one temperature only but F_i is not known (yet):

$$D(E) = h_i(E)e^{\beta_i(E-F_i)} \quad (2)$$

- We will average $D(E)$ from several simulations at different temperatures.
 - $D(E)$ does not depend on T_i , but our calculation does \pm statistical errors.
 - At different T_i different ranges of E are sampled.
- \Rightarrow We compose the total $D(E)$ from all data:

$$D(E) = \sum_i w_i(E) h_i(E) e^{\beta_i(E-F_i)}, \quad \sum_i w_i(E) = 1$$

Multiple histogram reweighting III

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Determining the weights: minimization of the error $\delta D(E)$ (or some estimate). Using reasonable assumptions, we get

$$w_i(E) = \frac{N_i h_i(E)}{\sum_j N_j h_j(E)} = \frac{N_i e^{-\beta_i(E-F_i)}}{\sum_j N_j e^{-\beta_j(E-F_j)}}$$

where N_i is the number of measurements at temperature β_i . \Rightarrow

$$e^{-\beta_i F_i} = \int D(E) e^{-\beta_i E} dE = \int \sum_j w_j(E) h_j(E) e^{\beta_j(E-F_j)} e^{-\beta_i E} dE$$

$$= \int \frac{\sum_j N_j e^{-\beta_j(E-F_j)} h_j(E) e^{\beta_j(E-F_j)}}{\sum_j N_j e^{-\beta_j(E-F_j)}} e^{-\beta_i E} dE = \int \frac{\sum_j N_j h_j(E) e^{-\beta_j E}}{\sum_j N_j e^{-\beta_j(E-F_j)}} dE$$

can be solved by iterations (self-consistent solution).

F_i are determined but an additive constant, $D(E)$ but a multiplicative factor

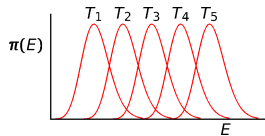
Multiple histogram reweighting IV

+ 23/37
s07/4

Expectation value at temperature β :

$$\langle X \rangle_\beta = \frac{\int X(E) D(E) e^{-\beta E} dE}{\int D(E) e^{-\beta E} dE} = \frac{\int X(E) \frac{\sum_i h_i(E)}{\sum_j N_j e^{-\beta_j(E-F_j)}} e^{-\beta E} dE}{\int \frac{\sum_i h_i(E)}{\sum_j N_j e^{-\beta_j(E-F_j)}} e^{-\beta E} dE}$$

- $\int dE$ is over histograms of width ΔE
- if ΔE is very short, all calculated energies are stored and $\int X(E) h_i(E) dE$ is replaced by a sum of E over $h_i(E) = 1/N_i$
- $\sqrt{\text{Var}E}/\langle E \rangle \approx N^{-1/2} \Rightarrow \beta_i/\beta_{i+1} \approx 1 \pm N^{-1/2}$ (overlapping distributions)



Parallel tempering (replica exchange)

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k simulations at temperatures $\beta_1 < \beta_2 \dots$ are run in parallel.

Once a while, 2 subsystems β_i, β_j (normally $|i-j| = 1$) are interchanged, the acceptance probability is

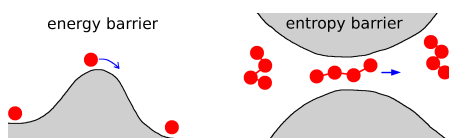
$$\min \left\{ 1, \frac{\exp(-\beta_j E_i - \beta_i E_j)}{\exp(-\beta_i E_i - \beta_j E_j)} \right\}$$

- Pros: easier barrier crossing, improved ergodicity, faster convergence at low temperatures
- Cons: correlations between subsystems, difficult error estimation

Conformational flooding, Wang-Landau, metadynamics

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Method suitable for fast **barrier crossing** (bad ergodicity - "bottleneck"), incl. calculation of the free energy profile (ΔF or ΔG), based on decreasing the energy barrier.



- (several predecessors)
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Conformational flooding, Wang-Landau, metadynamics

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● Our system is defined by the potential $U_0(r^N)$

● **Collective variable** $\lambda = \lambda(r^N)$ describes the studied process, e.g.: reaction coordinate, atom-atom distance, group-group distance (of their centers of mass e.g., ligand-receptor). various projections or angles...

There may be 2 or even more collective variables

● We simulate system with potential $U(r^N) = U_0(r^N) + \Delta U(\lambda)$, where $\Delta U = 0$ at start

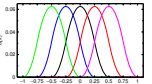
● $\Delta U(\lambda)$ is periodically updated: $\Delta U := \Delta U + \omega \frac{\delta(\lambda)}{h(\lambda)}$, $\lambda = \lambda(r^N)$

δ = approximation of δ -function (MC: histogram bin, better and MD: Gauss)

ω = small enough relaxation parameter, $\omega \ll k_B T$

$h(\lambda)$ = density of Cartesian points on hypersurface λ ; e.g., $h(\lambda) = 4\pi r_{12}^2$ for $\lambda = r_{12}$

\Rightarrow the probability of visiting the same λ again decreases



● If $h(\lambda)$ is omitted in the update stage, it has to be included later.

● Other weight functions may more efficient (= sample different ranges of λ better) than $h(\lambda)$; e.g., $h(\lambda) = r_{12}$ for $\lambda = r_{12}$. Again, the true density of points must be taken into account later.

Conformational flooding, Wang-Landau, metadynamics

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Free energy profile along λ :

● strictly, updating should be turned off in the final run (MC: microreversibility violation, MD: heating)

● then, the residual almost-uniform $p(\lambda)$ is determined and:

$$F(\lambda) = \text{const} - \Delta U(\lambda) - k_B T \ln[p(\lambda)/h(\lambda)]$$

● in practice, with small enough ω and continued updating, we can assume $p(\lambda)/h(\lambda) = \text{const}$ so that (in the sampled region of λ)

$$F(\lambda) = \text{const} - \Delta U(\lambda)$$

Free energy of a well-define "basin" of states (e.g., a bond state) is:

$$F = -k_B T \ln \int_{\lambda_1}^{\lambda_2} e^{-F(\lambda)/k_B T} h(\lambda) d\lambda$$

integration over region for more collective variables

! symbols const are different

Two case studies:

- simple MC (next 2 pages)
- potential of mean force in simen08

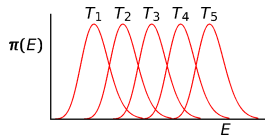
Multiple histogram reweighting IV

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Conformational flooding, Wang-Landau, metadynamics: Case study

plot/metadynamics.sh 1 2 28/37
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3D system, interaction energy:

online simulation is for $T=1$ K and $T=2$ K, type [r] to reread

$$\frac{U_0(x, y, z)}{k_B K} = \phi(x)(y^2 + z^2 + 1), \quad \text{where } \phi(x) = [(x+1)(x+2)(x-2)(x-4) + 21] e^{(x+1.5)^2/3}$$

- The barrier is $E^*/k_B = 1910$ K, $e^{-E^*/k_B T} = 10^{-929}$
- Collective variable $\lambda = x$
- Histogram: triangular $\delta(\lambda)$, grid = 1/100
- On purpose short MC step in x (\approx MD)
- Initial $\omega = 0.25 k_B T$, decrease to finish
- fine + stop ($\omega \equiv \alpha < 0$): aaaV in the plot window

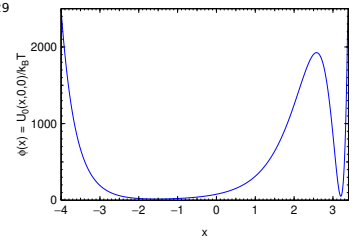
● graphs shown:

- running: $\Delta U - F_{\text{exact}}$, where

$$F_{\text{exact}}(\lambda) = \phi(\lambda) + k_B T \ln(\phi(\lambda))$$

- final: ΔU and $\phi(x) = U_0(x, 0, 0)$

- F and F_{exact} comparison



Note: in real systems, the second term will be a result of many degrees of freedom, not just 2 (y, z)

Parallel tempering (replica exchange)

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s07/4

k simulations at temperatures $\beta_1 < \beta_2 \dots$ are run in parallel.

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Conformational flooding, Wang-Landau, metadynamics: Case study

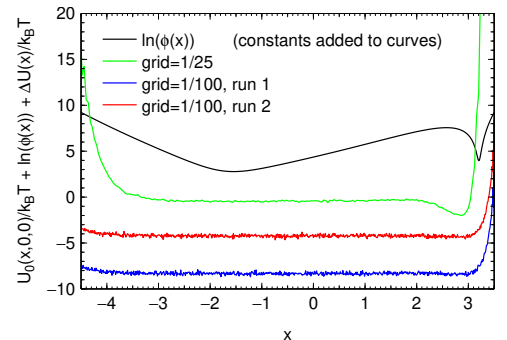
xmple_/_maple/metadynamics-case-study.mw 29/37
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The final graph shows $F(x) + \Delta U(x)$ (two independent simulations $T = 1$ K, different start)

1/grid	$\Delta F/k_B T$
50	37.85(8)
100	40.52(5)
200	40.98(5)
400	41.08(3)
exact*	41.13

where ΔF is the difference between the right and left basins.

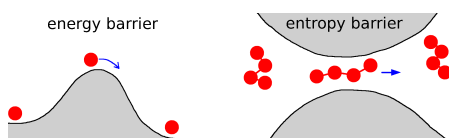
* see Maple



Conformational flooding, Wang-Landau, metadynamics

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Method suitable for fast **barrier crossing** (bad ergodicity - "bottleneck"), incl. calculation of the free energy profile (ΔF or ΔG), based on decreasing the energy barrier.



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Widom particle insertion method I

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Open system

$$dF = -SdT - pdV + \mu dN$$

$$\beta\mu = \left(\frac{\partial \beta F}{\partial N} \right)_{V,T} = - \left(\frac{\partial \ln Z_N}{\partial N} \right)_{V,T}$$

$$\beta\mu_{\text{res}} = \left(\frac{\partial \beta F_{\text{res}}}{\partial N} \right)_{V,T} = - \left(\frac{\partial \ln(Q_N/V^M)}{\partial N} \right)_{V,T} \approx - \left(\ln \frac{Q_{N+1}}{V^{M+1}} - \ln \frac{Q_N}{V^M} \right)$$

$$\exp(-\beta\mu_{\text{res}}) = \frac{1}{V} \frac{Q_{N+1}}{Q_N}$$

Or for the full chemical potential:

$$e^{-\beta\mu} = \frac{Z_{N+1}}{Z_N} = \frac{Q_{N+1}/[(N+1)! \Lambda^{3(N+1)}]}{Q_N/[N! \Lambda^{3N}]} = \frac{1}{(N+1)\Lambda^3} \frac{Q_{N+1}}{Q_N} \approx \frac{1}{N\Lambda^3} \frac{Q_{N+1}}{Q_N}$$

then by subtracting $\mu^{\text{id}} = k_B T \ln\left(\frac{N\Lambda^3}{V}\right)$ we get the same $\mu_{\text{res}} = \mu - \mu^{\text{id}}$

Widom particle insertion method II

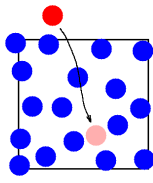
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$$\exp(-\beta\mu_{res}) = \frac{1}{V} \frac{Q_{N+1}}{Q_N}$$

$$N \rightarrow N+1 \quad U_{N+1} = U_N + \Psi(N)$$

$$\frac{1}{V} \frac{Q_{N+1}}{Q_N} = \frac{1}{V Q_N} \int \exp(-\beta U_N - \beta \Psi) d\vec{r}_{N+1} \dots d\vec{r}_{N+1} = \frac{1}{V} \int (e^{-\beta \Psi})_N d\vec{r}_{N+1}$$

$$\exp(-\beta\mu_{res}) = \frac{1}{V} \int (e^{-\beta \Psi})_N d\vec{r}_{N+1} = \langle (e^{-\beta \Psi})_N \rangle_{\text{random } \vec{r}_{N+1}}$$



where $\frac{1}{V} \int X d\vec{r}_{N+1} = \langle X \rangle_{\text{random } \vec{r}_{N+1}}$ = mean value of X over positions of the $(N+1)$ -th particle in volume V , calculated by MC integration (inserting a particle at random place)

- $(N+1)$ -th particle does not influence the system – it is virtual (fictitious, ghost)
- Problem: dense systems, large solutes
- Remedy: gradual insertion (thermodynamic integration or by finite steps)

Similar: solute insertion \Rightarrow solubility, Henry constant

Example

cd ../maple; xmaple simul07+18+Widom.mw 32/37
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We have simulated $N = 500$ Ar atoms (Lennard-Jones: $\sigma = 3.405 \text{ \AA}$, $\epsilon/k_B = 119.8 \text{ K}$) in a box of volume $V = 15.791 \text{ nm}^3$ at temperature $T = 150 \text{ K}$. By the Widom method, we found that[§]

$$\exp(-\mu_{res}/k_B T) = 0.749(3)$$

Calculate μ^o , the chemical potential of Ar with respect to the standard state of ideal gas at pressure $p^{\text{st}} = 1 \text{ bar}$ and temperature T .

Hints:

$$\mu_{res} = \mu - \mu^{\text{id}}(T, V) \quad (N \text{ particles in volume } V)$$

$$\mu^o = \mu - \mu^{\text{id}}(T, V^{\text{id}}) \quad (V^{\text{id}} = \text{volume of ideal gas of } N \text{ particles at } T, p^{\text{st}})$$

$$\mu^o = \mu_{res} + \mu^{\text{id}}(T, V) - \mu^{\text{id}}(T, V^{\text{id}})$$

$$= \mu_{res} + k_B T \ln \frac{N k_B T}{p^{\text{st}} V}$$

$$\mu^{\text{id}}(V) = k_B T \ln \frac{N \Lambda^3}{V}$$

[§]The value in parentheses is the estimated standard uncertainty in the unit of the least significant digit.

Henry constant

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Gas (2) dissolved in liquid (1), mole fraction in liquid $x_2 = N_2/(N_1 + N_2)$, $x_2 \ll 1$. One form of the Henry law for partial pressure p_2 of (2) in equilibrium with solution:

$$p_2 = K_H x_2$$

By virtual insertion of one molecule ($N_2 = 1$) of gas (2) to pure liquid (1), we get the residual chemical potential of (2) at $x_2 = 1/(N_1 + 1) \approx 1/N_1$,

$$\mu_{res,2} = \mu_2(x_2) - \mu_2^{\text{id}}(V)$$

In equilibrium:

$$\mu_2(x_2) \stackrel{\text{eq!}}{=} \mu^{\text{id}}(V_2) = k_B T \ln \frac{\Lambda^3}{V_2}$$

$$\mu_{res,2} = k_B T \ln \frac{N_2 \Lambda^3}{V_2} - k_B T \ln \frac{N_2 \Lambda^3}{V} = k_B T \ln \frac{V}{V_2}$$

where V_2 is the volume corresponding to pressure p_2 (don't forget that $N_2 = 1$),

$$V_2 = \frac{1 k_B T}{p_2} = \frac{k_B T}{x_2 K_H} = \frac{k_B T N_1}{K_H}$$

where $\rho_1 = N_1/V$ = number density of liquid (1). Finally,

$$K_H = \frac{\rho_1 k_B T}{e^{-\beta \mu_{res,2}}} = \frac{\rho_1 k_B T}{\langle (e^{-\beta \Psi})_{N_1} \rangle_{\text{random insertion of (2)}}$$

Reversible work by integrating the mean force

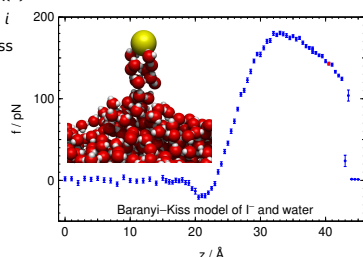
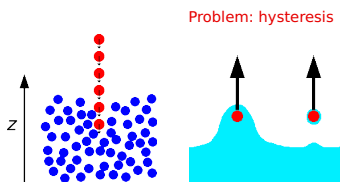
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From thermodynamics:

$$\Delta G = W_{\text{other than pressure-volume}} [\rho, T]$$

$$\Delta \mu_i = - \int_{\vec{r}_i(1)}^{\vec{r}_i(2)} \langle \vec{f}_i \rangle \cdot d\vec{r}_i$$

where $\vec{f}_i = -\partial U / \partial \vec{r}_i$ is the force acting on particle i
Molecules: the force applies to the center of mass or other reference point.



Mean force and its potential

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Let us define the **single-particle density** as (also denoted as n_1)

$$\rho_1(\vec{r}_1) = \frac{N}{Q_N} \int e^{-\beta U(\vec{r}_1, \dots, \vec{r}_N)} d\vec{r}_2 \dots d\vec{r}_N$$

– $\rho_1(\vec{r}_1) d\vec{r}_1$ is the probability of finding a particle (any one) in $d\vec{r}_1$
– extension to mixtures: $N/Q_N \rightarrow N_{\text{species}}/Q_N$

The **potential of mean force** is defined by

$$\bar{U}_1(\vec{r}_1) = -k_B T \ln[V \rho_1(\vec{r}_1)]$$

The corresponding force is

$$\vec{f}_i = - \left(\frac{\partial \bar{U}_1}{\partial \vec{r}_1} \right) = k_B T \frac{\partial \rho_1 / \partial \vec{r}_1}{\rho_1} = \left\langle - \left(\frac{\partial U}{\partial \vec{r}_1} \right) \right\rangle_{\vec{r}_2, \dots, \vec{r}_N} = \langle \vec{f}_i \rangle_{\vec{r}_2, \dots, \vec{r}_N}$$

i.e., it is indeed the mean force on particle 1 held at fixed position \vec{r}_1 .

Note: Similarly, for a 2-particle distribution, which for a pair of particles in an isotropic fluid is $g(r)$, one defines the potential of mean force as:

$$\bar{U}_2(r) = -k_B T \ln[g(r)]$$

Interfacial (surface) energy of solids

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Cleaving [Davidchack, Laird: JCP 118, 7651 (2003)]: Thermodynamic integration over a gradually growing “knife” (e.g., Gaussian potential) inserted between crystallographic planes.

Molding [Espinosa, Vega, Sanz: JCP 141, 134709 (2014)]: Thermodynamic integration over a gradually growing “mold” (potential wells) to keep a crystal growing in a part of the system.

- General problem of both methods: hysteresis

Gibbs energy of crystals

+

● Einstein crystal[†] at given T as a reference, integration over a coupling parameter [Frenkel, Ladd: JCP 81 3188 (1984), Frenkel, Mulder: Mol. Phys. 55, 1171 (1985)].

● Classical method: harmonic vibrations as a reference, thermodynamic integration of the difference $0 \rightarrow T$ [Kolafa JCTC 15, 68 (2019) and references therein]

[†]independent harmonic oscillators, here treated classically

Local density/concentration method

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Let a solute i be a subject of external potential $U_i^{\text{ext}}(\vec{r})$ (e.g., “gravity”). In equilibrium:

$$\mu_i(\vec{r}) + U_i^{\text{ext}}(\vec{r}) = \text{const}$$

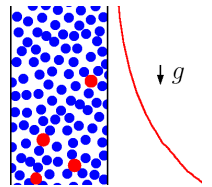
or

$$\mu_i(\vec{r}_1) - \mu_i(\vec{r}_2) = -[U_i^{\text{ext}}(\vec{r}_1) - U_i^{\text{ext}}(\vec{r}_2)]$$

We determine the concentration or density at \vec{r}_1 and know the chemical potential (with respect to a certain reference)

Example:

- reference = infinite dilution approximation ($\gamma = 1$) in the region of small concentration
- the activity coefficient γ in the region of high concentration can be calculated



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s07/4

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