3. patro 3rd floo Department of Physical Chemistry University of Chemistry and Technology, Prague Technická 5 (building A) http://www.mapy.cz/s/98vC jiri.kolafa@vscht.cz 220 444 257 Google: Kolafa Molecular modeling and simulation Kolafa Molekulární modelování a simulace

Molecular simulations

molecular dynamics (MD)

time development of a system composed of many molecules

instantaneous forces acting on atoms cause their motion

Monte Carlo (MC); more precisely: Metropolis method and its variants

 a sequence of configurations of the system is generated using random numbers an attempt is made to move a molecule (randomly with a certain distribution)

this trial move is accepted or rejected so that its probability is the same as in reality

kinetic Monte Carlo

the simulated process is divided into elementary events (e.g., adsorption of an atom on a growing crystal, catalytical reaction)

the event to happen is chosen according to a known probability

quantum simulations – MD, MC

Las Vegas algorithms - deterministic result (random pivot, search for minimum )

## **Elements of modeling**

Info

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? elementary particles + gravity, GUT, dark energy, . . .

known elementary particles: Standard model, atomic nuclei,...

Nuclei + electrons + photons: QED, accurate spectroscopy

 Nuclei + electrons: Schrödinger equation - small molecules, spectra, gas-phase equilibria, chemical kinetic, photochemistry ...

Atoms\* – classical (or quantum) atomistic modeling

Coarsed-grained models: meso/nanoscopic scale element = polyatomic group  $(surfactant = head + tail, polymer = [bead]_{n...})$ 

Microscopic scale (dispersions, granular materials)

Ontinuum: partial differential equations – heat, weather models, neutrons in A-bomb, statics in civil engineering

gravity: spacetime – black holes, gravitational waves

multiscale modeling: QM/MM (enzymes etc.) ... \*optionally: auxiliary interaction centers (sites), bigger groups (-CH<sub>3</sub>)

how structure affects properties (anomalies of water), solutions

phase equilibria, solubility

What can be studied

surfaces, interfaces, surfactants

Solids:

crystal structure, materials (defects)

adsorption (zeolites)

Biochemistry:

oproteins, nucleic acids, ion channels, lipid membranes

micelles, polymers, self-assembly (coarse-grained models, lattices)

Similar methods can be used for:

granular materials, optimization, spreading of epidemics, active matter, agent-based models, evo-devo (evolutionary developmental biology).

Structure optimization (example of molecular mechanics)

## **Potential Energy Surface (PES)**

Nuclei are much heavier than electrons ⇒ electron motion is much faster (Born-Oppenheimer approximation)

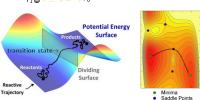
Potential energy surface (PES):

### energy as a function of positions of all nuclei

 $E_{\text{pot}}(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_N)$ 

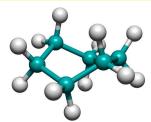
Chemical example: reaction coordinate

A reaction proceeds over (near to) a sad dle point (transition state)



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chair experiment: 28 kJ/mol model: 26 kJ/mol



twist (skew) boat experiment: 45 kJ/mol model: 53 kJ/mol

### How to obtain PES?

from quantum calculations (Schrödinger equation; ab initio, DFT).

approximated by a formula ("force field", "potential", "model", ...) 

combination: QM/MM methods (quantum mechanics/molecular mechanics)

### PES and modeling in chemistry

using classical mechanics: static calculations: minimum of energy, potential around a molecule

 time development of system (molecular dynamics, MD); with ab initio PES = AIMD (demanding!) thermodynamic variables by sampling (Monte Carlo, MC)

using quantum mechanics to nuclei:

- path integral methods (PI MC, PI MD); with AIMD extremely demanding

- classical mechanics + quantum corrections

combination force field + classical mechanics = "molecular mechanics" (MM); strictly speaking does not include MC and MD

# Molecular mechanics: static approach

We use PES, usually described by a force field

 $\bigcirc$  Energy minimization (T=0)

'molecular mechanics", "structure optimization'

Refinement – more accurate structure (from diffraction data) QSAR (Quantitative Structure–Activity Relationship) descriptors

Biochemistry: molecular shape (lock and key), hydrophillic/hydrophobic

.. but what about motion?

# Liquid water (example of equilibrium molecular dynamics)

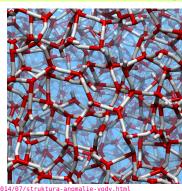
10000 molecules

300 K

periodic in x, y

adhesive pad

nonadhesive lid

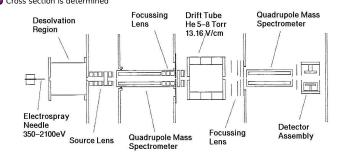


More movies: https://vesmir.cz/cz/on-line-clanky/20

### Electrospray of Cytochrome C

Electrospray: spray of charged aerosol

Cross section is determined



Yi Mao, J. Woenckhaus, J. Kolafa, M. A. Ratner, M. F. Jarrold: J. Am. Chem. Soc. 121, 2712-2721 (1999)

