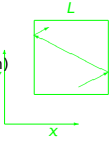


**Statistical thermodynamics instant** [jvk pic/BoltzmannTomb.jpg] 1/14 s01/3

- monoatomic ideal gas in a box:
  - ideal gas equation of state
  - temperature given by kinetic energy,  $\langle \frac{1}{2} m \bar{v}_{i,x}^2 \rangle = \frac{1}{2} k_B T$  (equipartition theorem) can be extended to any classical mechanical system
- ergodic hypothesis in the microcanonical ensemble:
  - quantum: all eigenstates have the same probability
  - classical: density of phase-space states is constant on the constant-energy manifold
- $\pi \propto \exp(-E/k_B T)$  in the canonical ensemble is derived by:
  - set of heat-exchanging sub-ensembles = big microcanonical ensemble
  - multiplication of prob. of two noninteracting systems,  $\pi(E_1 + E_2) = \pi(E_1)\pi(E_2)$
- Boltzmann equation for entropy
 
$$dU = \sum_{\psi} \pi(\psi) \cdot d\mathcal{E}(\psi) + \sum_{\psi} d\pi(\psi) \cdot \mathcal{E}(\psi) = -p dV + T dS$$

$$\Rightarrow S = -k_B \sum_{\psi} \pi(\psi) \ln \pi(\psi) \text{ or } S = k_B \ln W \Rightarrow F, G, \text{ etc.}$$
- Boltzmann H-theorem (second law)



**Semiclassical monoatomic ideal gas** 6/14 s01/3

$$Q = \int \exp[0] d\vec{r}_1 \dots d\vec{r}_N = \int_V d\vec{r}_1 \dots \int_V d\vec{r}_N = V^N$$

$$Z = \frac{Q}{N! \Lambda^{3N}} = \frac{V^N}{N! \Lambda^{3N}} \approx \frac{V^N}{N^N e^{-N} \Lambda^{3N}}, \quad F = -k_B T \ln Z = -k_B T N \ln \frac{Ve}{N \Lambda^3}$$

$$p = -\left(\frac{\partial F}{\partial V}\right)_T = \frac{k_B T N}{V} = \frac{nRT}{V} \quad \text{e = Euler number}$$

$$U = F + TS = F - T \left(\frac{\partial F}{\partial T}\right)_V = \frac{3Nk_B T}{2} \quad \text{e = elementary charge}$$

$$\mu = \left(\frac{\partial F}{\partial N}\right)_{T,V} = k_B T \ln \left(\frac{N \Lambda^3}{V}\right) = k_B T \ln \left(\frac{\rho \Lambda^3}{k_B T}\right)$$

(with respect to the standard state of a free molecule at zero temperature)  
And verification:

$$G = F + pV = k_B T N \ln \frac{N \Lambda^3}{Ve} + Nk_B T = N\mu$$

**Canonical ensemble: Helmholtz energy** 2/14 s01/3

$$\pi(\psi) = \frac{e^{-\beta \mathcal{E}(\psi)}}{Z}, \quad Z = \sum_{\psi} e^{-\beta \mathcal{E}(\psi)} \quad \beta = \frac{1}{k_B T}$$

$$S = -k_B \sum_{\psi} \pi(\psi) \ln \pi(\psi) = -k_B \sum_{\psi} \pi(\psi) [-\beta \mathcal{E}(\psi) - \ln Z] = \frac{U}{T} + k_B \ln Z$$

$\Rightarrow$  Helmholtz energy:  $F = -k_B T \ln Z$

$Z =$  canonical partition function = statistical sum (also denoted  $Q$ )  
Interpretation: number of "accessible" states (low-energy states are easily accessible, high-energy states are not)

From the Helmholtz energy  $F$  we can obtain all quantities ( $dF = -pdV - SdT$ ):

$$p = -\frac{\partial F}{\partial V} \quad U = F + TS$$

$$S = -\frac{\partial F}{\partial T} \quad H = U + pV$$

$$G = F + pV$$

**Monoatomic ideal gas** 7/14 s01/3

Or quantum calculation of the translational partition function:  
Eigenvalues of energy of a point mass in a  $a \times b \times c$  box:

$$\mathcal{E} = \frac{h^2}{8m} \left( \frac{n_x^2}{a^2} + \frac{n_y^2}{b^2} + \frac{n_z^2}{c^2} \right)$$

**Maxwell-Boltzmann statistics:** high enough temperature so that a few particles compete for the same quantum state – it does not matter whether we have fermions or bosons; equivalently,  $\Lambda \ll$  distance between particles.

**Partition function:**

$$Z_1 = \sum_{n_x=1}^{\infty} \sum_{n_y=1}^{\infty} \sum_{n_z=1}^{\infty} \exp(-\beta \mathcal{E}) \approx \int_0^{\infty} \int_0^{\infty} \int_0^{\infty} \exp(-\beta \mathcal{E}) dn_x dn_y dn_z = \frac{V}{\Lambda^3}$$

$$E = \sum_{i=1}^N \mathcal{E}_i \Rightarrow Z = \frac{1}{N!} Z_1^N$$

**Yes, it is the same!** The choice of factor  $1/h^{3N}$  in the semiclassical  $Z$  was correct.

**Semiclassical partition function** 3/14 s01/3

Hamilton formalism: positions of atoms =  $\vec{r}_i$ , momenta =  $\vec{p}_i$ .

$$\mathcal{E} = \mathcal{H} = E_{\text{pot}} + E_{\text{kin}}, \quad E_{\text{pot}} = U(\vec{r}_1, \dots, \vec{r}_N), \quad E_{\text{kin}} = \sum_i \frac{\vec{p}_i^2}{2m}$$

Sum over states replaced by integrals (classical mechanics needed):

$$Z = \sum_{\psi} e^{-\beta \mathcal{E}(\psi)} = \frac{1}{N! h^{3N}} \int \exp[-\beta H(\vec{r}_1, \vec{p}_1, \dots, \vec{r}_N, \vec{p}_1, \dots, \vec{p}_N)] d\vec{r}_1 \dots d\vec{p}_N$$

where  $h = 2\pi\hbar$  = Planck constant.

**Why the factorial?**

- Particles are indistinguishable ... but at high enough  $T$  appear in different quantum states

**Why Planck constant?**

- Has the correct dimension ( $Z$  must be dimensionless)
- We get the same result for noninteracting quantum particles in a box

**but** fails if quantum effects are important (vide infra)

**Polyatomic ideal gas** 8/14 s01/3

The internal degrees of freedom are separated from the translational ones.  
The internal partition function is the sum over the internal degrees of freedom:

$$q = \sum_{\psi_{\text{in}}} e^{-\beta \mathcal{E}(\psi)}$$

Canonical partition function:

$$Z = \frac{(qV)^N}{N! \Lambda^{3N}}, \quad F = -k_B T \ln Z = -k_B T N \ln \frac{Ve q}{N \Lambda^3}$$

Chemical potential:

$$\mu = \left(\frac{\partial F}{\partial N}\right)_{T,V} = k_B T \ln \left(\frac{N \Lambda^3}{qV}\right) = k_B T \ln \left(\frac{\rho \Lambda^3}{q k_B T}\right)$$

**Semiclassical partition function** 4/14 s01/3

**Integrals over positions and momenta are separated**

Integrals over momenta can be evaluated:  $\int \exp\left(-\frac{p_{1,x}^2/2m}{k_B T}\right) dp_{1,x} = \sqrt{2\pi m k_B T}$  After  $3N$  integrations we get:

$$Z = \frac{Q}{N! \Lambda^{3N}}, \quad \text{de Broglie thermal wavelength: } \Lambda = \frac{h}{\sqrt{2\pi m k_B T}}$$

$\Lambda =$  de Broglie wavelength at typical particle velocity at given  $T$   
requirement:  $\Lambda \ll$  typical atom-atom separation  $\approx (V/N)^{1/3}$

Configurational integral:

$$Q = \int \exp[-\beta U(\vec{r}_1, \dots, \vec{r}_N)] d\vec{r}_1 \dots d\vec{r}_N$$

do not confuse:  
 $U$  = internal energy  
 $U(\vec{r}_1, \dots) =$  potential

Mean value of a **static** quantity (observable):

$$\langle X \rangle = \frac{1}{Q} \int X(\vec{r}_1, \dots, \vec{r}_N) \exp[-\beta U(\vec{r}_1, \dots, \vec{r}_N)] d\vec{r}_1 \dots d\vec{r}_N$$

**NPT ensemble:  $T, p = \text{const}$**  9/14 s01/3

Also isothermal-isobaric, loosely "isobaric":  $V \rightarrow p$   
The same argument now applied to  $V$  (as for  $E$  before):

$$\pi(V_{1+2}) = \pi(V_1 + V_2) = \pi(V_1) \pi(V_2)$$

Together:  $\pi = \exp(\alpha_i - \beta E - \gamma V)$   
 $\gamma$  is a universal property of a barostat – to be determined from ideal gas

$$\langle V \rangle = \frac{\int V e^{-\beta E_{\text{kin}} - \gamma V} dV d\vec{r}_1 \dots d\vec{p}_N}{\int e^{-\beta E_{\text{kin}} - \gamma V} dV d\vec{r}_1 \dots d\vec{p}_N} = \frac{\int V^{N+1} e^{-\gamma V} dV}{\int V^N e^{-\gamma V} dV} = \frac{N+1}{\gamma}$$

Tricks used:

- $\int d\vec{p}_1 \dots d\vec{p}_N$  gives the same factor,  $\Lambda^{-3N}$
- $\int d\vec{r}_1 \dots d\vec{r}_N = V^N$  correctly:  $\int X dV d\vec{r}_1 \dots d\vec{r}_N = \int_0^{\infty} [ \int_V X d\vec{r}_1 \dots d\vec{r}_N ] dV$  (the order of integration is important –  $\int dV$  is the last) or by substituting  $V^{1/3} \xi_i = \vec{r}_i$  one gets  $d\vec{r}_1 \dots d\vec{r}_N = V^N d\xi_1 \dots d\xi_N$  (then the integration order is irrelevant)
- $\int_0^{\infty} V^N e^{-\gamma V} dV = N!/\gamma^{N+1}$  (recursively by parts)

( $V$ ) should be equal to  $Nk_B T/p$  (at limit  $N \rightarrow \infty$ )  $\Rightarrow \gamma = \frac{N+1}{N} \frac{p}{k_B T} \approx \frac{p}{k_B T}$

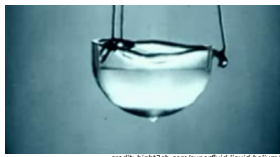
For nitpickers: (more in simen10):  
 $(V_{NPT}(p, N, T)) - V \approx \frac{k_B T}{2N} \left(\frac{\partial p}{\partial T}\right) \frac{\rho^2 \partial^2 V}{\partial \rho^2}$   
id. gas  $\frac{V}{N} = \frac{p}{k_B T}$   
so that +1 below can be safely ignored...

**Thermal de Broglie wavelength** 5/14 s01/3

**Example**

a) Calculate  $\Lambda$  for helium at  $T = 2$  K.  
b) Compare to the typical distance of atoms in liquid helium (density 0.125 g/cm<sup>3</sup>).

a)

$$\Lambda = \frac{h}{\sqrt{2\pi m k_B T}} = \frac{6.6 \times 10^{-34}}{\sqrt{2 \times \pi \times \frac{0.004}{6 \times 10^{23}} \times 1.38 \times 10^{-23} \times 2}} = 6.2 \times 10^{-10} \text{ m}$$


b)

$$l = \sqrt[3]{V_1} = \sqrt[3]{\frac{M}{N \rho}} = \sqrt[3]{\frac{0.004}{6 \times 10^{23} \times 125}} = 3.8 \times 10^{-10} \text{ m}$$

$l < \Lambda \Rightarrow$  cannot use classical mechanics

**NPT ensemble (contd.)** 10/14 s01/3

Normalization constant  $\alpha$  (from the Boltzmann eq. and  $e^{\alpha} = 1/Z_{NPT}$ ):

$$S = k_B \sum_{\psi} \pi(\psi) [\alpha - \beta \mathcal{E}(\psi) - \gamma V] = k_B \alpha - \frac{U}{T} - \frac{p(V)}{T}$$

$$-k_B T \ln Z_{NPT} = U - TS + p(V) = G$$

$$Z_{NPT} = \frac{1}{N! h^{3N}} \int \chi e^{-\beta(E+pV)} d\vec{r}_1 \dots d\vec{p}_N dV$$

where  $\chi = \{1/\Lambda^3, \beta p, N/V, 1/V, \dots\}$  zajišťuje bezrozměrnost ( $N \rightarrow \infty$  stejné)

We easily get:  $dG = -SdT + Vdp$

$$\left(\frac{\partial \ln Z_{NPT}}{\partial p}\right)_T = -\beta(V) \frac{N+1}{N} + \frac{1}{p} \text{ čili } \left(\frac{\partial G}{\partial p}\right)_T = (V) \frac{N+1}{N} - \frac{k_B T}{p} \xrightarrow{N \rightarrow \infty} (V)$$

The expectation value of  $X$  in the isobaric ensemble is

$$\langle X \rangle = \frac{\int \chi e^{-\beta(E+pV)} dV d\vec{r}_1 \dots d\vec{p}_N}{Z_{NPT}}$$

**Grandcanonical ensemble:  $\mu = \text{const}$**

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The same argument now applied to  $N$  (as for  $E, V$  before):

$$\pi(N_{1+2}) = \pi(N_1 + N_2) = \pi(N_1)\pi(N_2)$$

Together:  $\pi = \exp(\alpha_i - \beta E + \delta N)$

$\delta$  is a universal property of a source of particles – to be determined from ideal gas

$$\langle N \rangle = \frac{\sum_{N=0}^{\infty} \int \frac{N}{h^{3N} N!} e^{-\beta E_{\text{kin}}} e^{\delta N} d\vec{r}_1 \dots d\vec{p}_N}{\sum_{N=0}^{\infty} \int \frac{1}{h^{3N} N!} e^{-\beta E_{\text{kin}}} e^{\delta N} d\vec{r}_1 \dots d\vec{p}_N} \stackrel{\text{id.}}{=} \frac{\sum_{N=0}^{\infty} \frac{N}{N!} \frac{V^N}{\Lambda^{3N}} e^{\delta N}}{\sum_{N=0}^{\infty} \frac{1}{N!} \frac{V^N}{\Lambda^{3N}} e^{\delta N}} = \frac{V}{\Lambda^3} e^{\delta}$$

Tricks used:

- $\int d\vec{p}_1 \dots d\vec{p}_N = \Lambda^{-3N}$
- $\int d\vec{r}_1 \dots d\vec{r}_N = V^N$
- $\sum_{N=0}^{\infty} \frac{1}{N!} x^N = e^x$ , where  $x = \frac{V^N}{\Lambda^{3N}} e^{\delta N}$
- derivative by  $x$ :  $\sum_{N=0}^{\infty} \frac{1}{N!} N x^{N-1} = e^x \Rightarrow \sum_{N=0}^{\infty} \frac{N}{N!} x^N = x e^x$

On comparing with  $\exp(-\beta \mu_{\text{id}}, \text{point particle}) = V/\Lambda^3$  one gets  $\delta = \beta \mu$ .

**Grandcanonical ensemble:  $\mu = \text{const}$  (contd.)**

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s01/3

Normalization constant  $\alpha$  (from the Boltzmann eq. and  $e^\alpha = 1/Z_{\mu VT}$ ):

$$S = k_B \sum_{\psi} \pi(\psi) [\alpha - \beta \mathcal{E}(\psi) + \delta N] = k_B \alpha - \frac{U}{T} + \frac{\mu \langle N \rangle}{T}$$

$$-k_B T \ln Z_{\mu VT} = U - TS - \mu \langle N \rangle = \Omega$$

where

$$Z_{\mu VT} = \sum_{N=0}^{\infty} \frac{e^{\beta \mu N}}{N! h^{3N}} \int e^{-\beta E} d\vec{r}_1 \dots d\vec{p}_N$$

**Grandcanonical potential**  $\Omega = F - \mu N = F - G = -pV$

$$dF = -SdT - pdV + \mu dN \Rightarrow d\Omega = -SdT - pdV - Nd\mu$$

**Grandcanonical ensemble:  $\mu = \text{const}$  (contd.)**

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The expectation value of  $X$  in the grandcanonical ensemble is

$$\langle X \rangle = \frac{\sum_0^\infty \frac{\alpha^N}{N!} \int X e^{-\beta E} d\vec{r}_1 \dots d\vec{p}_N}{\sum_0^\infty \frac{\alpha^N}{N!} \int e^{-\beta E} d\vec{r}_1 \dots d\vec{p}_N}, \text{ where } \alpha = e^{\beta \mu}.$$

The last eq. is usually integrated over momenta (incl. internal degrees of freedom); for  $X = X(N, \vec{r}^N)$  it holds:

$$\langle X \rangle = \frac{\sum_0^\infty \frac{\alpha'^N}{N!} \int X e^{-\beta E} d\vec{r}_1 \dots d\vec{r}_N}{\sum_0^\infty \frac{\alpha'^N}{N!} \int e^{-\beta E} d\vec{r}_1 \dots d\vec{r}_N}, \text{ where } \alpha' = \frac{\alpha q}{\Lambda^3} \quad (1)$$

**Grandcanonical ensemble:  $\mu = \text{const}$  (contd.)**

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The quotient in series (1) can be expressed as

$$e^{\beta \mu_{\text{id}}} = \rho \Lambda^3 / q$$

$$\alpha' = \frac{\alpha q}{\Lambda^3} = e^{\beta \mu_{\text{res}}} \rho, \quad \rho = \frac{\langle N \rangle}{V}, \quad \mu_{\text{res}} = \mu - \mu^{\text{id}}(\rho)$$

where  $\mu_{\text{res}}$  is the residual chemical potential = chemical potential with respect to the standard state of ideal gas at given temperature and volume (= density), which can be compared with tables (after pressure is recalculated from  $\rho^{\text{st}}$  using the ideal gas equation of state).